

## Assignment 5 — Solutions [*Revision* : 1.3]

**Q10.1** The equation of hydrostatic equilibrium is

$$\frac{dP}{dr} = -g\rho.$$

Recalling that the definition of optical depth is

$$d\tau = -\bar{\kappa}\rho dr,$$

we can rewrite the HE equation as

$$-\bar{\kappa}\rho \frac{dP}{d\tau} = -g\rho.$$

Rearranging,

$$\frac{dP}{d\tau} = \frac{g}{\bar{\kappa}}$$

which is the desired result.

**Q10.3** If the Sun is composed completely of hydrogen atoms, then the number of atoms will be given by

$$N = \frac{M_{\odot}}{m_{\text{H}}}.$$

Plugging in the numbers,  $N = 1.2 \times 10^{57}$ .

If 10 eV is released by every atom, the total amount of energy liberated over the Sun's lifetime would be

$$E = N \cdot 10 \text{ eV} = 1.9 \times 10^{39} \text{ J}$$

The lifetime of the Sun, at its present luminosity, would therefore be

$$\tau = \frac{E}{L_{\odot}} = 5.0 \times 10^{12} \text{ s} = 158,000 \text{ yr}$$

This is much shorter than the age of the Earth (as determined from geology and paleontology), and therefore it is not possible that the Sun's energy is entirely chemical.

**Q10.4** (a). The height of the Coulomb barrier between two protons is

$$U \approx \frac{1}{4\pi\epsilon_0} \frac{e^2}{r}$$

where  $r$  is their separation. Assuming  $r \approx 10^{-15} \text{ m}$ , we find  $U \approx 1.44 \text{ MeV} \approx 2.31 \times 10^{-13} \text{ J}$ .

If this barrier can be penetrated by protons with a velocity  $v = 10 v_{\text{rms}}$ , where  $v_{\text{rms}}$  is the root-mean-square velocity of the Maxwell-Boltzmann distribution, then these penetrating photons must have a kinetic energy

$$E \geq 10^2 \cdot \frac{3kT}{2}.$$

(The fractional term on the right-hand side is the kinetic energy at  $v = v_{\text{rms}}$ ). Equating this with the barrier height derived above,

$$100 \frac{3kT}{2} \geq 2.31 \times 10^{-13} \text{ J}.$$

Solving for the temperature, we obtain  $T = 1.11 \times 10^8 \text{ K}$ . This is approaching a factor of ten larger than the actual central temperature of the Sun,  $T_{\text{c}} \approx 1.6 \times 10^7 \text{ K}$ .

(b). With a little algebra, eqn. (8.1) of O&C can be written in the form

$$n_v dv = n \left( \frac{3}{2\pi v_{\text{rms}}} \right)^{3/2} e^{-3v^2/2v_{\text{rms}}^2} 4\pi \left( \frac{v}{v_{\text{rms}}} \right)^2 dv,$$

where  $v_{\text{rms}} \equiv \sqrt{3kT/m}$  is the RMS velocity. Hence, the ratio between the number of particles per unit velocity interval at  $v = v_{\text{rms}}$  and  $v = 10 v_{\text{rms}}$  is

$$\frac{n_{10v_{\text{rms}}}}{n_{v_{\text{rms}}}} = \frac{100e^{-150}}{e^{-1.5}} = 3.22 \times 10^{-63}$$

(c). From the previous question,  $N \approx 1.2 \times 10^{57}$ . Given the ratio calculated above, we can see that, on average, not even a single proton will be moving with a velocity of ten times  $v_{\text{rms}}$ ! Therefore, in the absence of tunneling there is no way that  $v = 10v_{\text{rms}}$  protons can account for the Sun's luminosity.

**Q10.5** The pressure integral is

$$P = \frac{1}{3} \int_0^\infty mn_v v^2 dv$$

(cf. O&C, eqn. 10.9). Substituting in the Maxwell-Boltzmann velocity distribution given by eqn. (8.1) of O&C, we have

$$P = \frac{1}{3} \int_0^\infty mn \left( \frac{m}{2\pi kT} \right)^{3/2} e^{-mv^2/2kT} 4\pi v^4 dv.$$

Making a change of variables to  $x = \sqrt{m/2kT}v$ , this can be rewritten as

$$P = \frac{1}{3} mn \left( \frac{m}{2\pi kT} \right)^{3/2} 4\pi \left( \frac{2kT}{m} \right)^{5/2} \int_0^\infty x^4 e^{-x^2} dx = \frac{8}{3\sqrt{\pi}} nkT \int_0^\infty x^4 e^{-x^2} dx$$

The definite integral evaluates (e.g., using Mathematica) to  $3\sqrt{\pi}/8$ . Hence, we obtain

$$P = nkT,$$

which is the ideal gas law.

**Q10.17** To prove that the  $n = 0$  polytrope has the solution

$$D_0(\xi) = 1 - \frac{\xi^2}{6},$$

we can substitute it into the Lane-Emden equation:

$$\frac{1}{\xi^2} \frac{d}{d\xi} \left[ \xi^2 \frac{dD_n}{d\xi} \right] = -D_n^n.$$

The left-hand side evaluates to

$$\frac{1}{\xi^2} \frac{d}{d\xi} \left[ \xi^2 \frac{dD_n}{d\xi} \right] = \frac{1}{\xi^2} \frac{d}{d\xi} \left[ \xi^2 \cdot -\frac{\xi}{3} \right] = -1$$

The right-hand side trivially evaluates to  $-1$  also (since  $x^0 = 1$  for any  $x \neq 0$ ). Hence, the solution above satisfies the Lane-Emden equation.

The remaining step is to verify that the solution also satisfies the boundary conditions. At  $\xi = 0$ , we have

$$D_0(0) = 1$$

and

$$D'_0(0) = 0,$$

which does indeed satisfy the conditions  $D_n(0) = 1$  and  $D'_n(0) = 0$ . Moreover,  $D_0(\xi) = 0$  at  $\xi = \xi_1 = \sqrt{6}$ .

**Q10.18** The density of a polytrope is given by

$$\rho = \rho_c D_n^n$$

(cf. p. 336 of O&C), where  $\rho_c$  is the central density and  $D_n$  the appropriate solution of the Lane-Emden equation. With  $n = 0$ , this becomes

$$\rho = \rho_c,$$

and so the  $n = 0$  polytrope can be recognized as having a uniform density structure. This polytrope is often referred to as the ‘homogeneous, incompressible’ case.

**Q10.20** (a). See Fig. 1 for the graph.

- (b). The curves indicate that the density is more centrally concentrated for increasing polytropic index. This statement can be placed on a more-rigorous footing by noting that the mean density of a polytrope is

$$\bar{\rho} = \frac{3M}{4\pi R^3} = -3\rho_c \xi_1^{-1} \left. \frac{dD_n}{d\xi} \right|_{\xi_1}$$

(using expressions given for the mass and radius by O&C). Hence, the ratio between the central density and  $\bar{\rho}$  (which measures the degree of central concentration) is

$$\frac{\rho_c}{\bar{\rho}} = \frac{1}{3} \left[ \xi_1^{-1} \cdot - \left. \frac{dD_n}{d\xi} \right|_{\xi_1} \right]^{-1}$$

Consulting any table of polytropes (or using *Poly-Web*) will show that both terms in the brackets are monotonically decreasing functions of  $n$ ; thus, the degree of central concentration will be a monotonically *increasing* function of  $n$ .

- (c). An adiabatically convective model has  $n = 1.5$ , whereas a model in radiative equilibrium has  $n = 3$ . Therefore, the latter should be more centrally concentrated than the former.
- (d). This question is too vague to answer.

**EZ-Web Polytrope Question** If the Sun’s structure is described by a polytrope, then the pressure and density should be related by

$$P = K\rho^{(n+1)/n}$$

for fixed  $K$  and  $n$ . Taking the logarithm of both sides gives

$$\log_{10} P = \frac{n+1}{n} \log_{10} \rho + \log_{10} K;$$

thus, a plot of the solar structure in the  $\log_{10} P - \log_{10} \rho$  plane can be used to determine the polytropic index.

Fig. 2 shows such a plot, for an *EZ-Web* model of the present-day Sun. The best fit line to the pressure/density data has a slope 1.41; hence, the polytropic index for the Sun is estimated from  $(n+1)/n \approx 1.41$ , as  $n \approx 2.44$ .

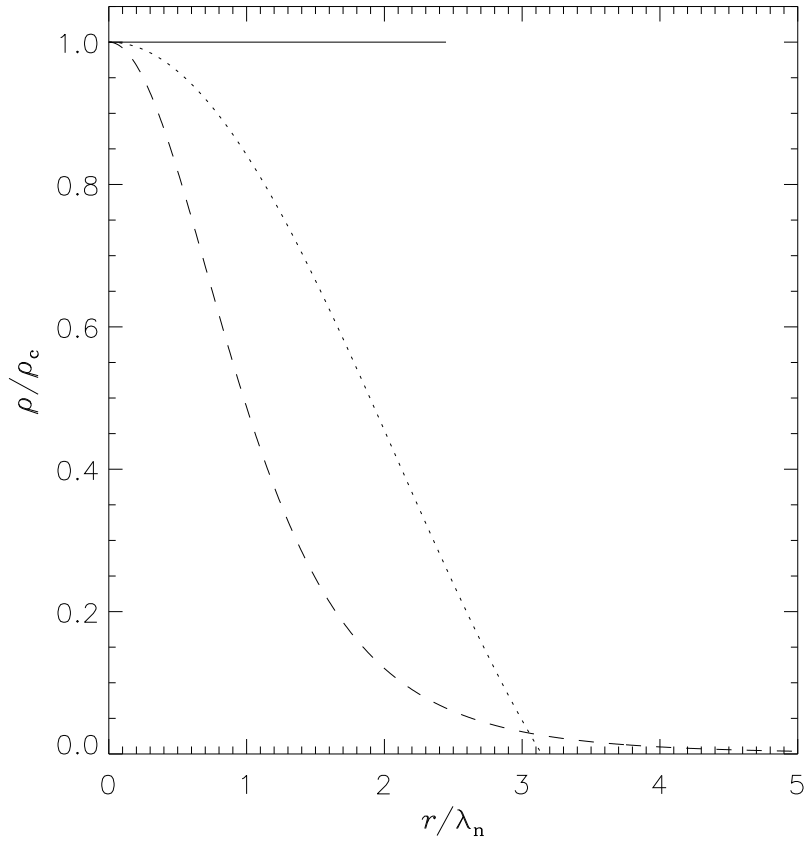


Figure 1: The scaled density  $\rho/\rho_c$  plotted as a function of scaled radius  $r/\lambda_n$  ( $= \xi$ ) for polytropes having indices  $n = 0$  (solid),  $n = 1$  (dotted) and  $n = 5$  (dashed). Note that the  $n = 0$  curve terminates at  $r/\lambda_n = \sqrt{6}$ , because this point corresponds to the stellar surface.

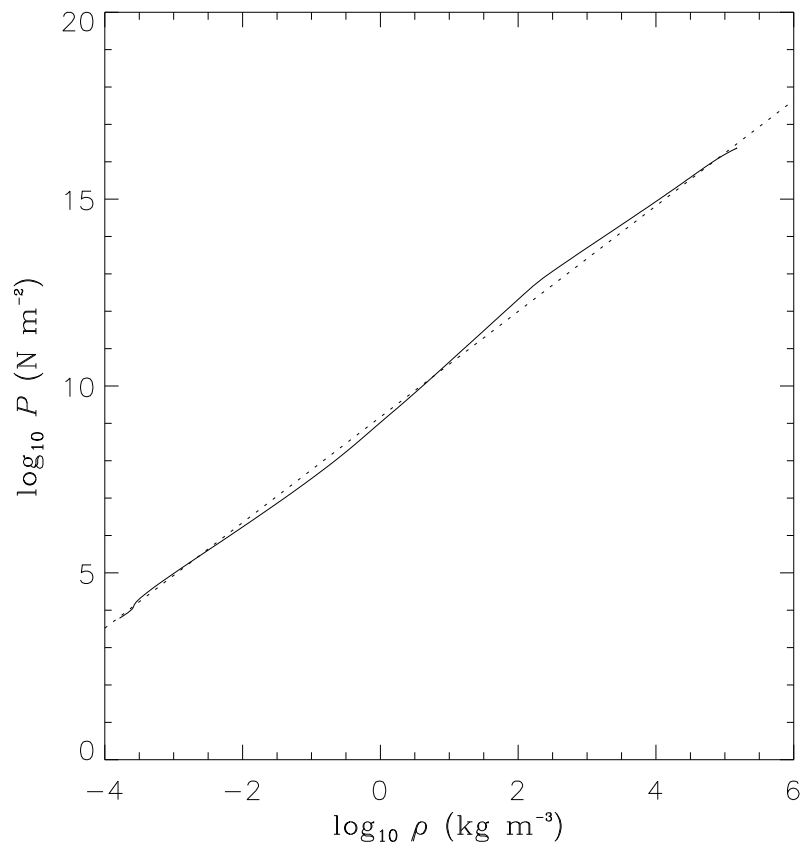


Figure 2: An *EZ-Web* model for the present-day Sun, plotted in the logarithmic pressure-density plane (solid). The dotted line shows the best fit to the *EZ-Web* data; it has a slope 1.41 and an intercept 9.17.