

27 — Nuclear Energy Generation [*Revision* : 1.2]

- Energy Sources

- Age of Moon from rocks: 4.5 billion years
- Recall from Notes 19: Kelvin-Helmholtz timescale

$$t_{\text{KH}} = \frac{GM^2}{LR}$$

- For Sun, $t_{\text{KH}} \sim 30 \text{ Myr}$, much shorter than age of Earth or solar system. So, the Sun cannot be powered by KH contraction
- Likewise, Sun cannot be powered by chemical energy, because energy release per atom is on order of $\sim 1 \text{ eV}$
- Only possible energy source is **nuclear**; recognized in early 20th Century, before details worked out

- Binding Energy

- Energy is released when nucleus is assembled from constituent nucleons; this is **binding energy** of nucleus
- Release comes from work done by **strong nuclear force**
- Express binding energy as

$$E_{\text{b}} = \Delta mc^2$$

- Δm is **mass defect** — difference in combined masses of constituent nucleons, m_j , and mass of nucleus m_{nuc} :

$$\Delta m = \sum_j m_j - m_{\text{nuc}}$$

- Classifying nuclei:
 - * Z is number of protons (**atomic number**)
 - * N is number of neutrons
 - * A is number of nucleons (protons + neutrons; **mass number**)
 - * $A = Z + N$
- Write mass of constituents as

$$\sum_j m_j = Zm_{\text{p}} + Nm_{\text{n}} = Zm_{\text{p}} + (A - Z)m_{\text{n}},$$

where m_{p} is proton mass, m_{n} is neutron mass

- So, binding energy is

$$E_{\text{b}} = [Zm_{\text{p}} + (A - Z)m_{\text{n}} - m_{\text{nuc}}] c^2$$

- Binding energy per nucleon is

$$\frac{E_{\text{b}}}{A} = \frac{[Zm_{\text{p}} + (A - Z)m_{\text{n}} - m_{\text{nuc}}] c^2}{A};$$

indication of how tightly bound each nucleon is, and so how much energy can be extracted through nuclear reactions

- If a reaction leads to net increase in E_{b}/A , then it is **exothermic**
- Graph of E_{b}/A against A is peaked at Fe ($\sim 8.5 \text{ MeV}$)

- For A smaller than 56, use **fusion** to release energy
- For A greater than 56, use **fission** to release energy

- **Thermonuclear Reactions**

- Nuclear reactions occur by collisions between nuclei and/or protons ('free' neutrons decay into protons in ~ 15 mins)
- To get particles close enough to react via short-range strong force, must overcome repulsion between positive charges
- Define close enough as radius of nucleus r_{nuc}
- At this distance, Coulomb (electrostatic) potential energy of two particles is

$$U_C = \frac{Z_1 Z_2 e^2}{4\pi\epsilon_0 r_{\text{nuc}}}$$

- This energy must come from kinetic energy of particles, and in turn from thermal energy. So, for **thermonuclear reactions**

$$\frac{3}{2}kT \approx U_C = \frac{Z_1 Z_2 e^2}{4\pi\epsilon_0 r_{\text{nuc}}}$$

- To find r_{nuc} , use de Broglie wavelength

$$r_{\text{nuc}} \sim \lambda \sim \frac{h}{p}$$

But

$$\frac{p^2}{2m} = \frac{3kT}{2}$$

and so

$$r_{\text{nuc}} = \frac{h}{\sqrt{3mkT}}$$

- Solving for T ,

$$T = \frac{Z_1^2 Z_2^2 e^4 m}{12\pi^2 \epsilon_0^2 h^2 k}$$

For proton-proton reaction, $Z_1 = Z_2 = 1$; but resulting T much higher than found e.g. in Sun

- Solution to paradox: **quantum-mechanical tunneling!**

- **Reaction rates**

- Consider reaction where target particle x interacts with incoming particles i
- Number of reactions per target in time dt :

$$N = \sigma v n_i dt$$

where σ is reaction cross section, v is velocity of incoming particles, and n_i is their number density

- **HOWEVER**, not all incoming particles have same energy; and σ strongly dependent on energy

- So, consider number of reactions per second per target with incoming particle in energy interval $(E, E + dE)$:

$$dN_E = \sigma(E)v(E)n_{i,E} dE dt$$

where $n_{i,E}$ is number of incoming particles with energy in same interval,

$$v(E) = \left(\frac{2E}{m_i}\right)^{1/2}$$

where m_i is incoming particle mass

- Assume incoming particles are in thermal equilibrium; number distribution given by Maxwell-Boltzmann distribution:

$$n_{i,E}dE = \frac{n_i}{n}n_EdE$$

where

$$n_E dE = \frac{2n}{\pi^{1/2}} \frac{1}{(kT)^{3/2}} E^{1/2} e^{-E/kT} dE$$

- Now must find energy-dependent cross-section $\sigma(E)$. Generally, complicated, but can at least get scalings
- For an actual interaction to take place, incoming particle must (a) ‘hit’ target particle, and (b) interact with it
- For (a), cross section of target varies with geometrical cross section:

$$\sigma(E) \propto \pi\lambda^2 \propto E^{-1}$$

where λ is de Broglie wavelength

- For (b), cross section of target varies with tunneling probability. This probability is exponential with height of Coulomb barrier:

$$\sigma(E) \propto e^{-2\pi^2 U_C/E}$$

(factor of $2\pi^2$ is from QM). Evaluate barrier height U_C from Coulomb potential at $r = \lambda$:

$$\frac{U_C}{E} = \frac{Z_1 Z_2 e^2}{4\pi\epsilon_0 \lambda E}$$

Final result:

$$\sigma(E) \propto e^{-bE^{-1/2}}$$

where

$$b = \frac{\pi m_i^{1/2} Z_1 Z_2 e^2}{2^{1/2} \epsilon_0 h}$$

- Putting (a) and (b) together:

$$\sigma(E) = \frac{S(E)}{E} e^{-bE^{-1/2}}$$

where function $S(E)$ is now some hopefully slowly-varying function of energy

- Reactions per second thus becomes

$$dN_E = \frac{S(E)}{E} e^{-bE^{-1/2}} \left(\frac{2E}{m_i}\right)^{1/2} \frac{2n}{\pi^{1/2}} \frac{1}{(kT)^{3/2}} E^{1/2} e^{-E/kT} dE dt$$

- Integrate over all energies, multiply by n_x divide by dt to get overall reaction rate per unit volume

$$r_{ix} = n_x \frac{\int dN_E}{dt} = \left(\frac{2}{kT} \right)^{3/2} \frac{n_i n_x}{(m_i \pi)^{1/2}} \int_0^\infty S(E) e^{-bE^{-1/2}} e^{-E/kT} dE.$$

- Integrand is product of rapidly dropping factor $e^{-E/kT}$ (due to decline in number of particles with high thermal energies), and rapidly growing factor $e^{-bE^{-1/2}}$ (due to increased probability of barrier penetration at high energies). Two factors combine to form strongly peaked curve — the **Gamow peak**
- Top of Gamow peak has energy

$$E_0 = \left(\frac{bkT}{2} \right)^{2/3}$$

Major contribution to rate integral over narrow energy interval centered on E_0 ; allows us to replace $S(E)$ with value at peak:

$$r_{ix} \approx n_x \frac{\int dN_E}{dt} = \left(\frac{2}{kT} \right)^{3/2} \frac{n_i n_x}{(m_i \pi)^{1/2}} S(E_0) \int_0^\infty e^{-bE^{-1/2}} e^{-E/kT} dE.$$

However, this assumes $S(E)$ is slowly-varying function of energy; in fact, sometimes sharp variations occur due to **resonances**

- One missing ingredient in above analysis: **electron screening** makes nuclei appear ‘less positive’, and makes reactions easier to occur
- Above expressions for reaction rate are complicated; often fit using simple power law

$$r_{ix} \approx r_0 X_i X_x \rho^{\alpha'} T^\beta$$

If each reaction creates energy \mathcal{E}_0 , energy generation rate per unit mass given by

$$\epsilon_{ix} = \frac{\mathcal{E}_0}{\rho} r_{ix} = \epsilon'_0 X_i X_x \rho^{\alpha'} T^\beta$$