

A cartoon illustration of a scientist with a large nose, glasses, and a wide, toothy grin. He is holding a test tube in his right hand and a bell in his left hand. The background is a teal color with some bubbles in the top left corner.

ASTRONOMY 103: THE EVOLVING UNIVERSE

Lecture 4

COSMIC CHEMISTRY

Substitute Lecturer: Paul Sell

Two Blackbody Trends

1. Wein's (Veen's) Law

$$\lambda_p \propto 1 / T$$

or

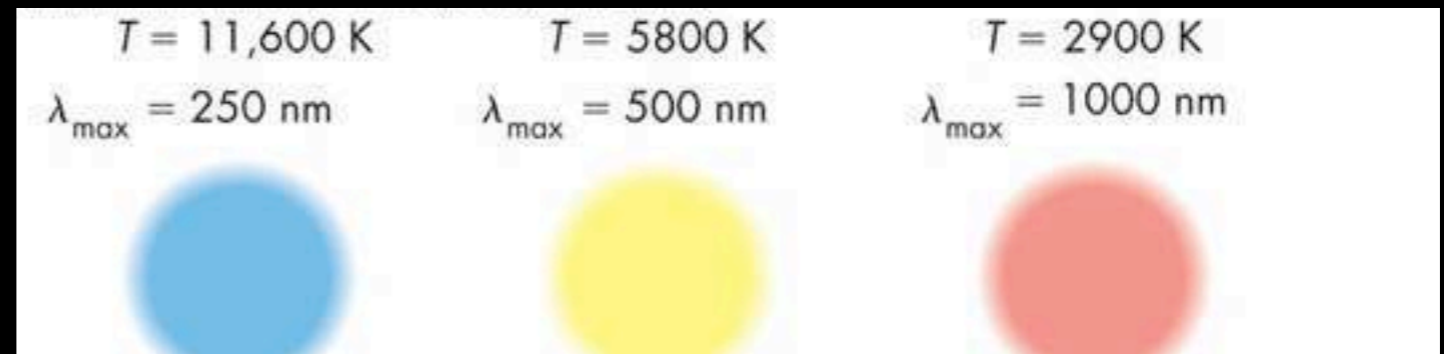
$$\lambda_p = 2900 / T$$

(λ_p is the "peak wavelength" in micrometers or microns and T is the temperature in Kelvin)

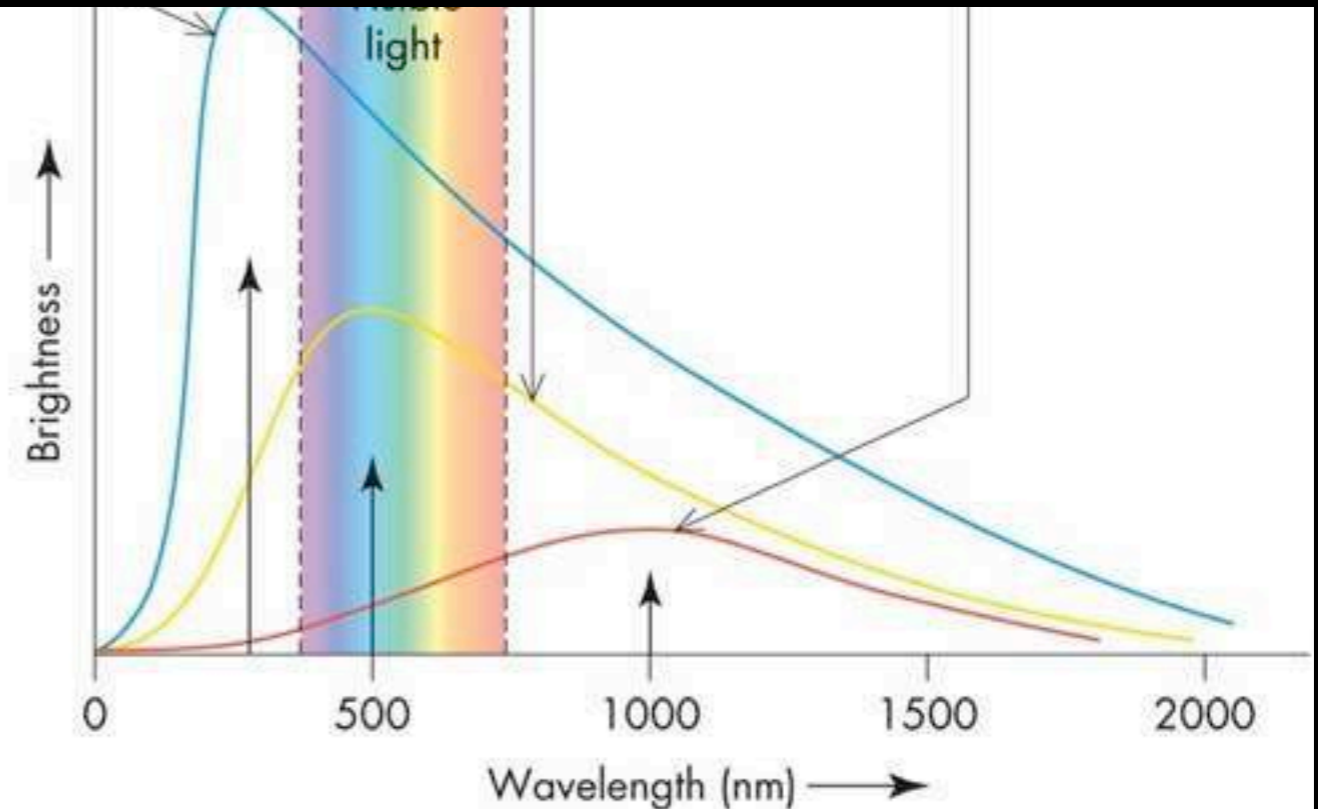
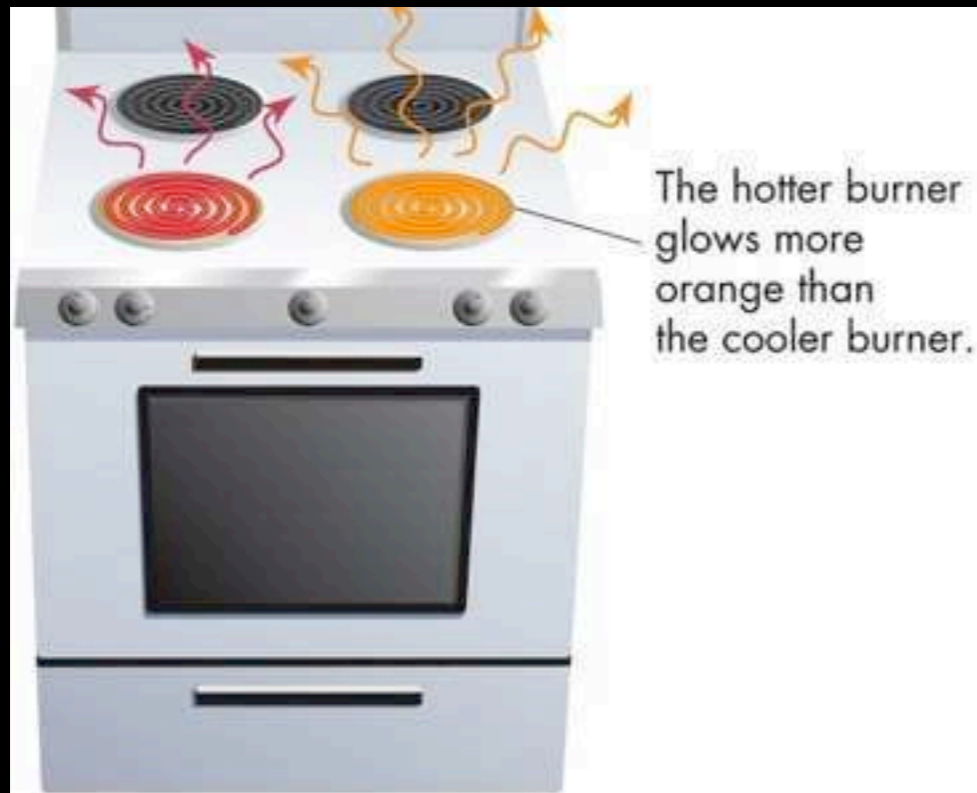
2. Stephan-Boltzman Law

$$F_{\text{tot}} \propto T^4 \quad \text{or} \quad F_{\text{tot}} = \sigma T^4$$

F_{tot} is the "total flux" or the amount of photons emitted per unit area from the surface of the object



How do you get a continuous or blackbody spectrum?



Multiple Choice Question

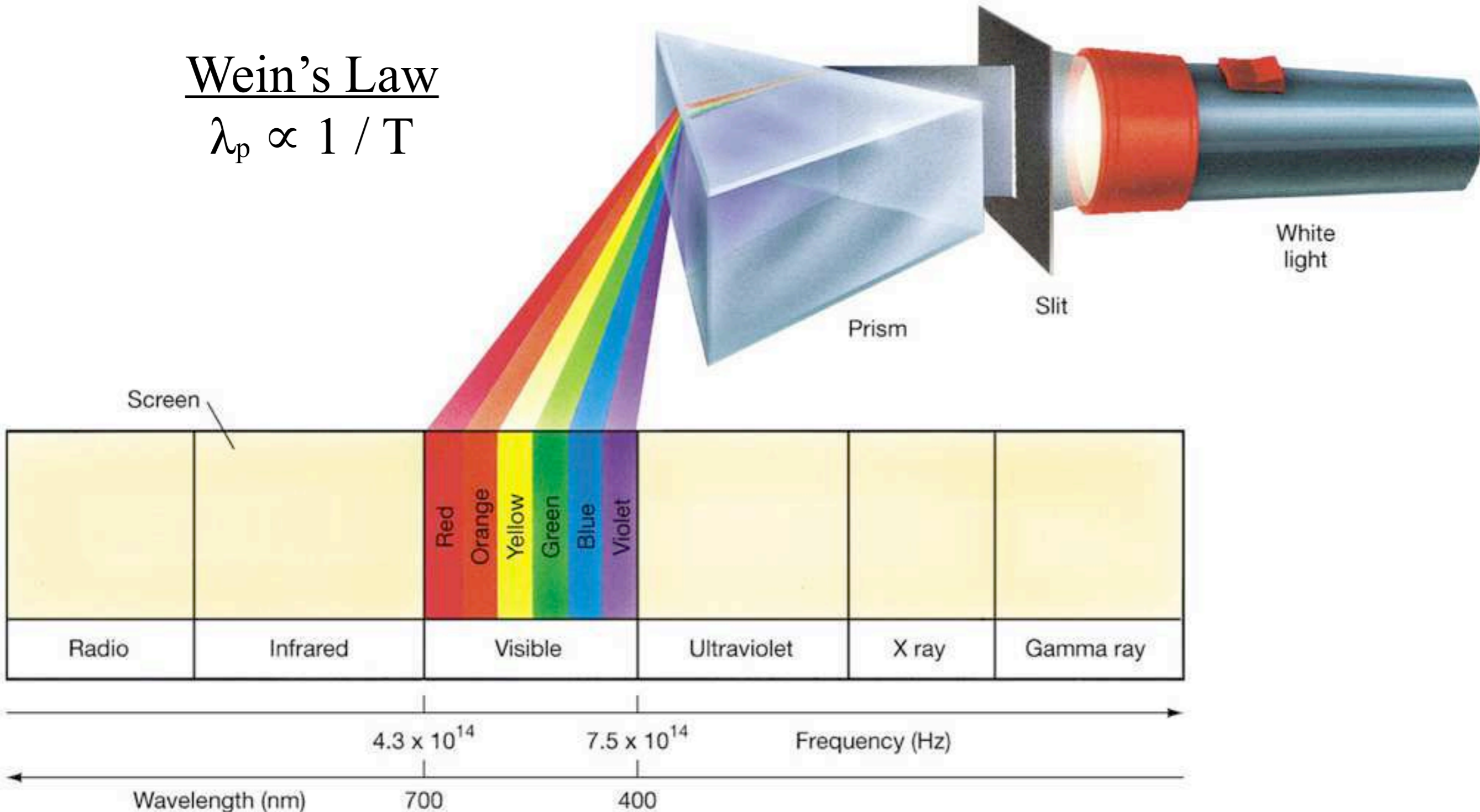
Q3.1 The sun's surface temperature is around 6000 degrees Kelvin. If it were to suddenly become hotter, where would most of its radiation be emitted?

- a) in the optical, ultraviolet, x-ray or gamma ray region, depending on the temperature change
- b) in the optical
- c) in the optical, infrared, microwave or radio region, depending on the temperature change
- d) uniformly at all wavelengths
- e) in the infrared where it is hottest

Multiple Choice Question

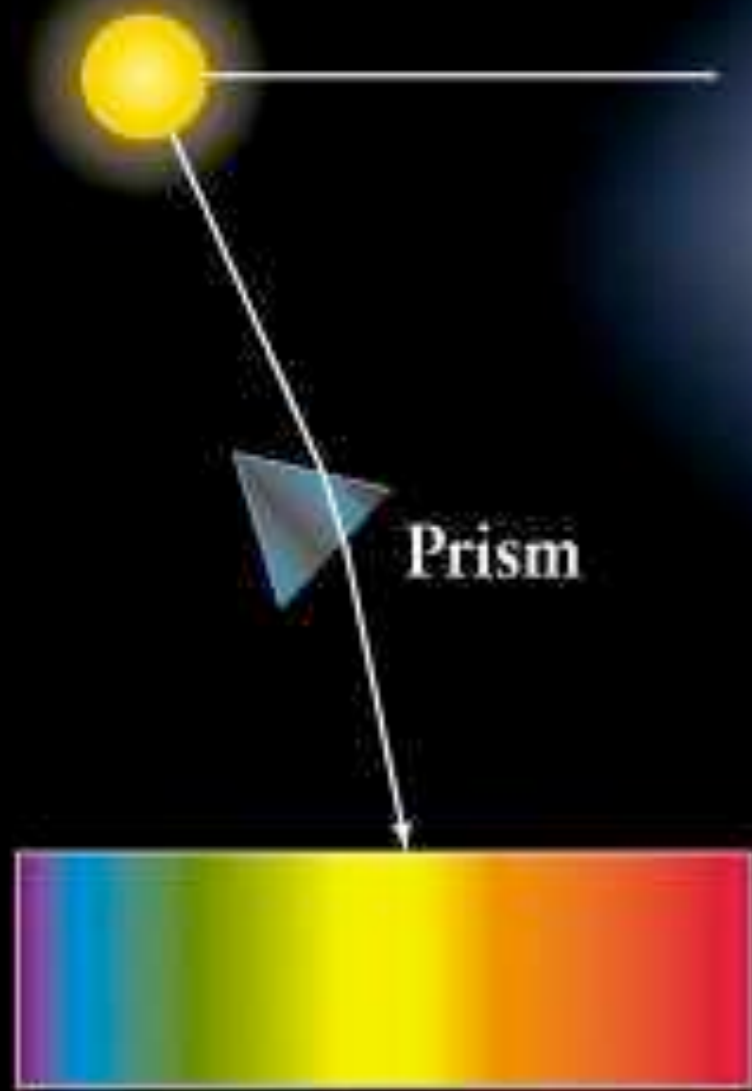
Wein's Law

$$\lambda_p \propto 1 / T$$



Kirchoff's Laws

Hot blackbody



a Continuous spectrum

Cloud of cooler gas

Prism



b Absorption line spectrum

3. A low-density gas absorbs light from a continuum source at the same discrete wavelengths. This produces absorption lines.



c Emission line spectrum

Emission:
light is added

Absorption:
light is removed

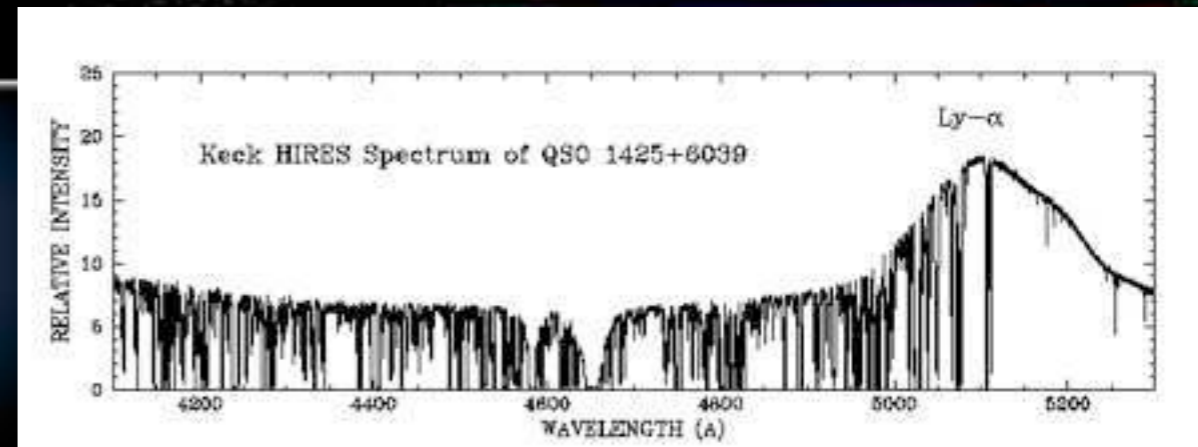
1. A luminous solid or liquid (or dense gas) emits light at all wavelengths. This produces a continuous spectrum of radiation.

2. A low-density gas emits light at discrete wavelengths (or frequencies) that are called emission lines.

Kirchoff's Laws

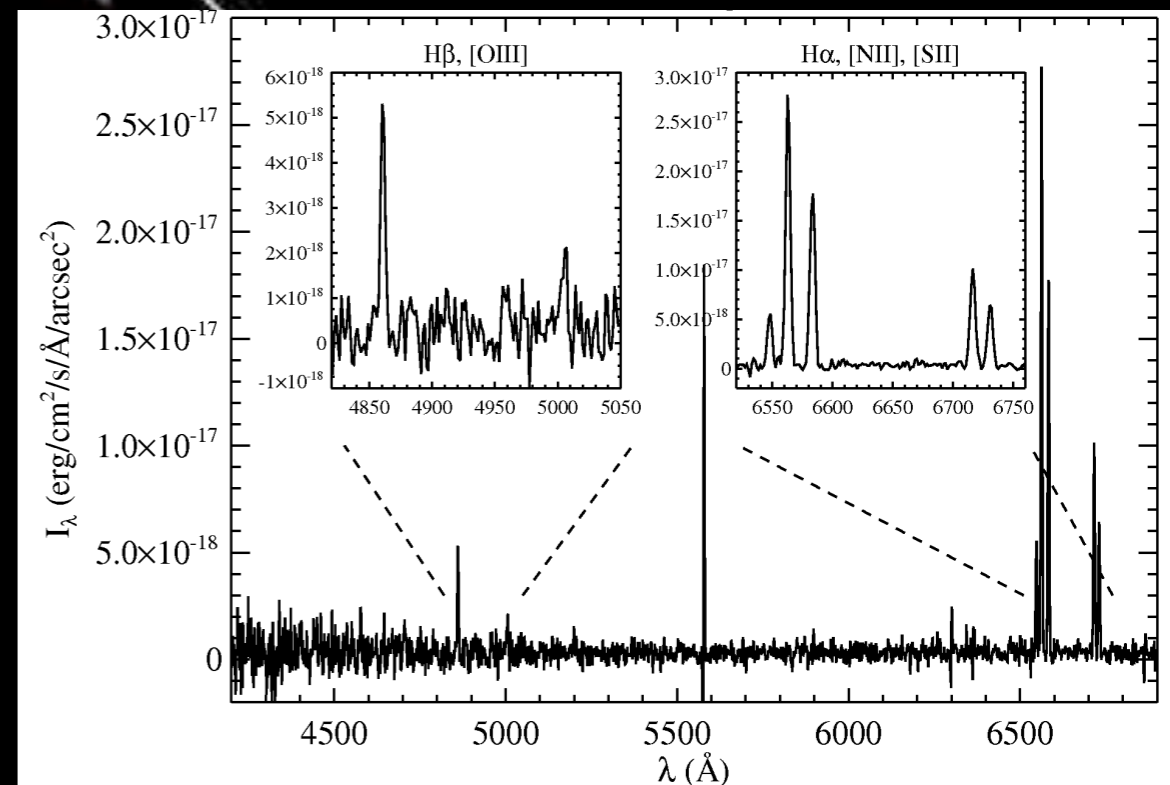
Absorption:
light is removed

Prism



light from a continuum source
at the same discrete
wavelengths. This produces
absorption lines.

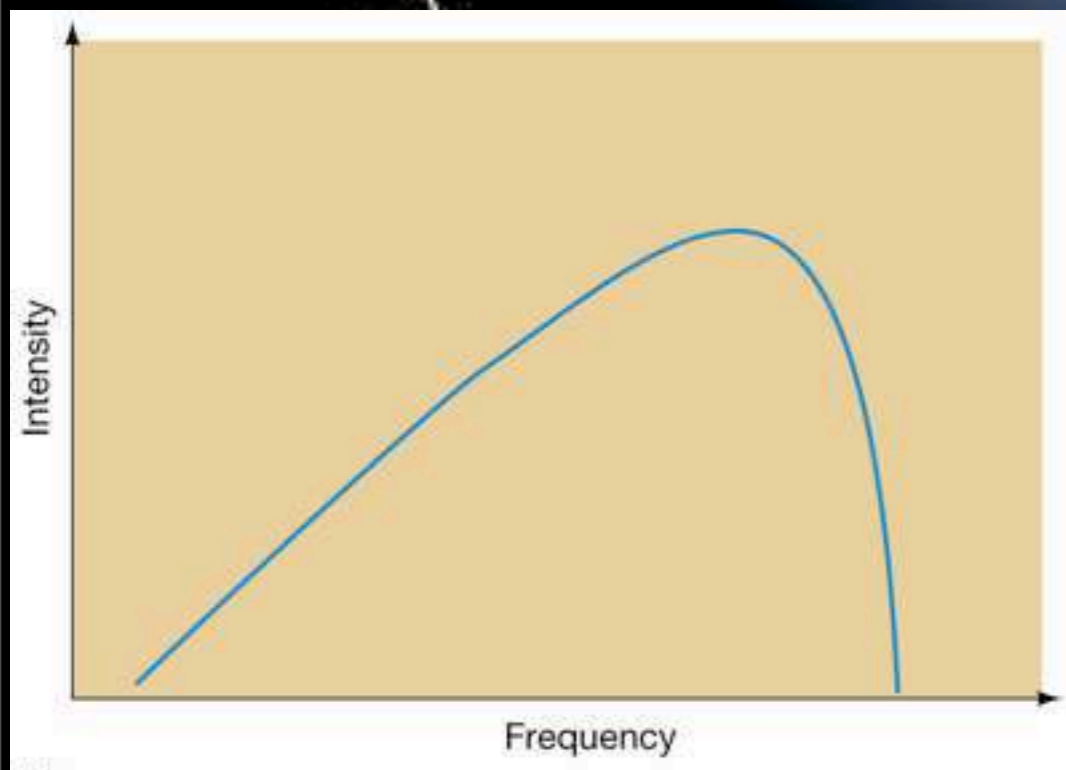
Emission:
light is added



Hot blackbody



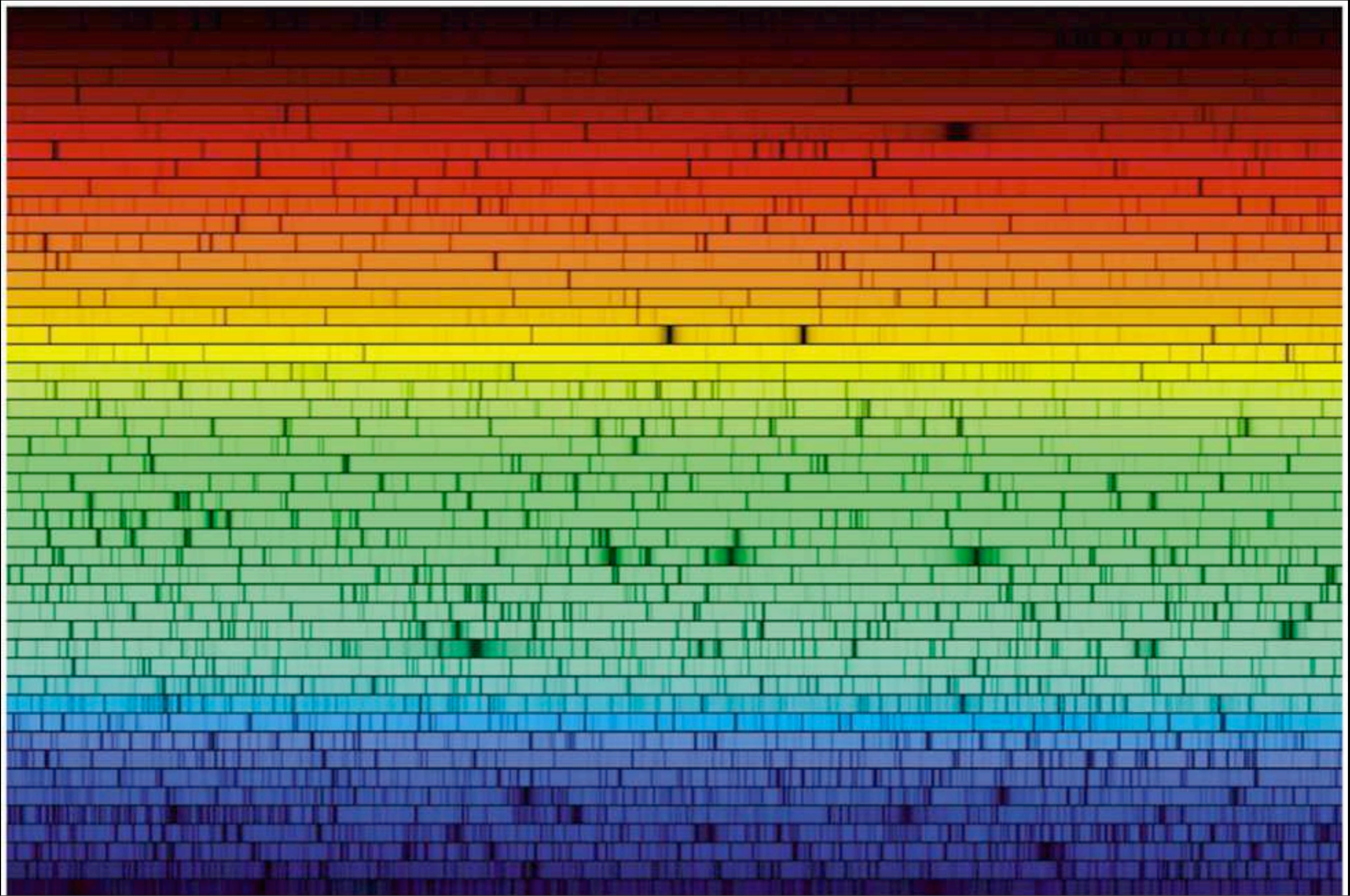
Cloud of cooler gas



1. A luminous solid or liquid (or dense gas) emits light at all wavelengths. This produces a continuous spectrum of radiation.

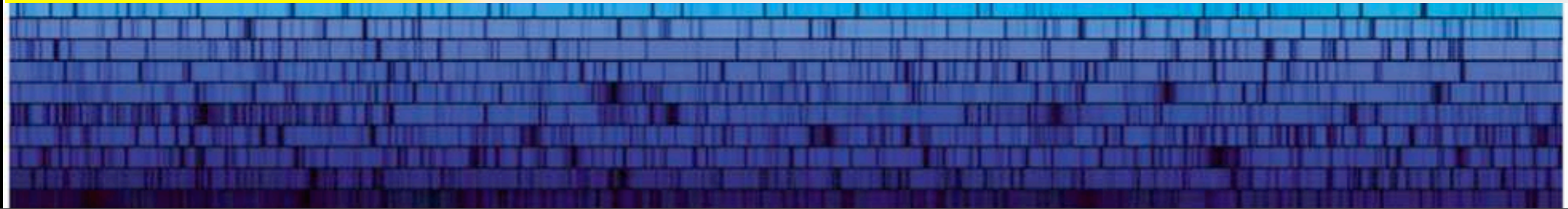
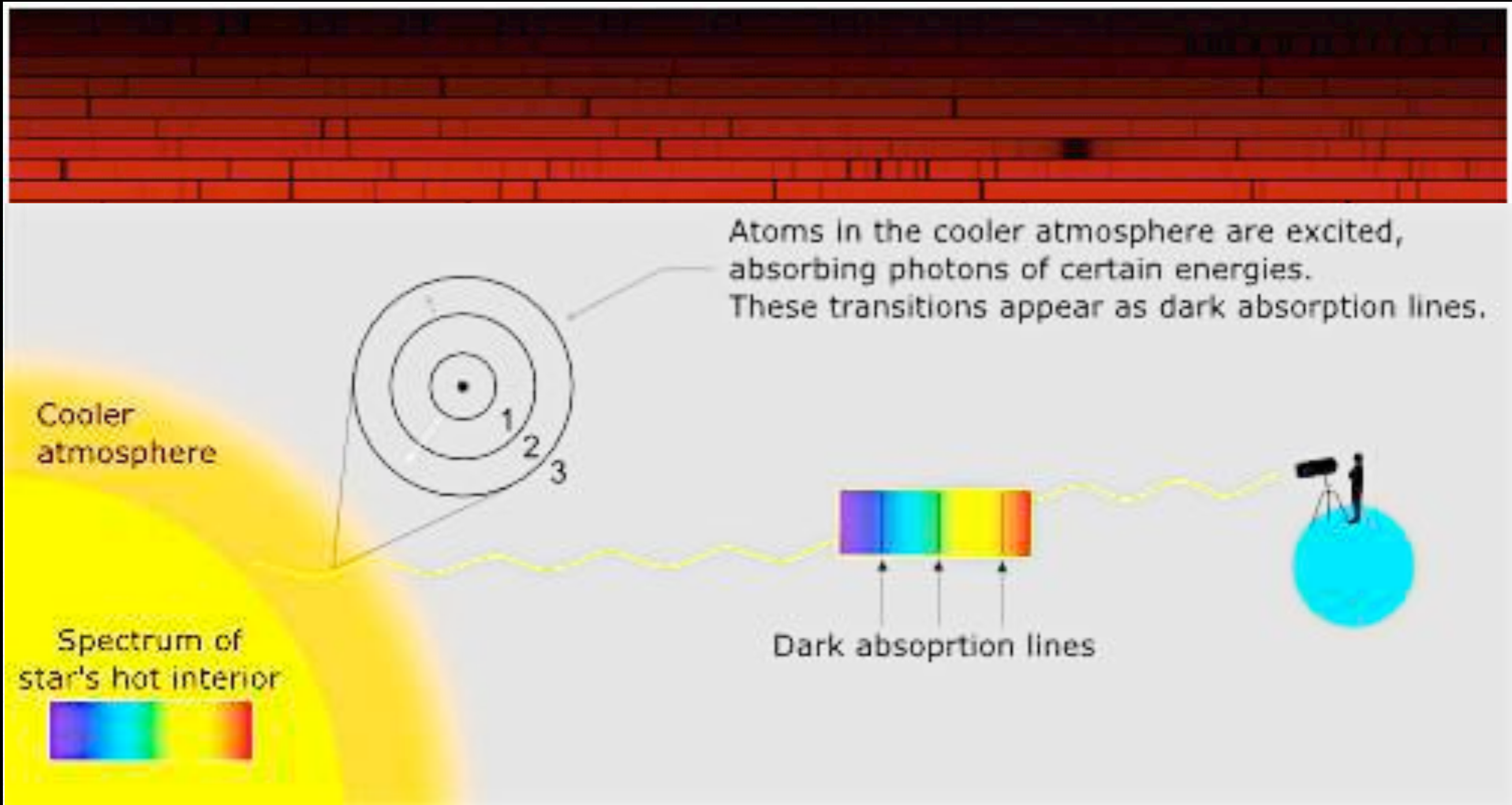
No Star is a Perfect Blackbody

Example: Fraunhofer Lines in the Solar Spectrum

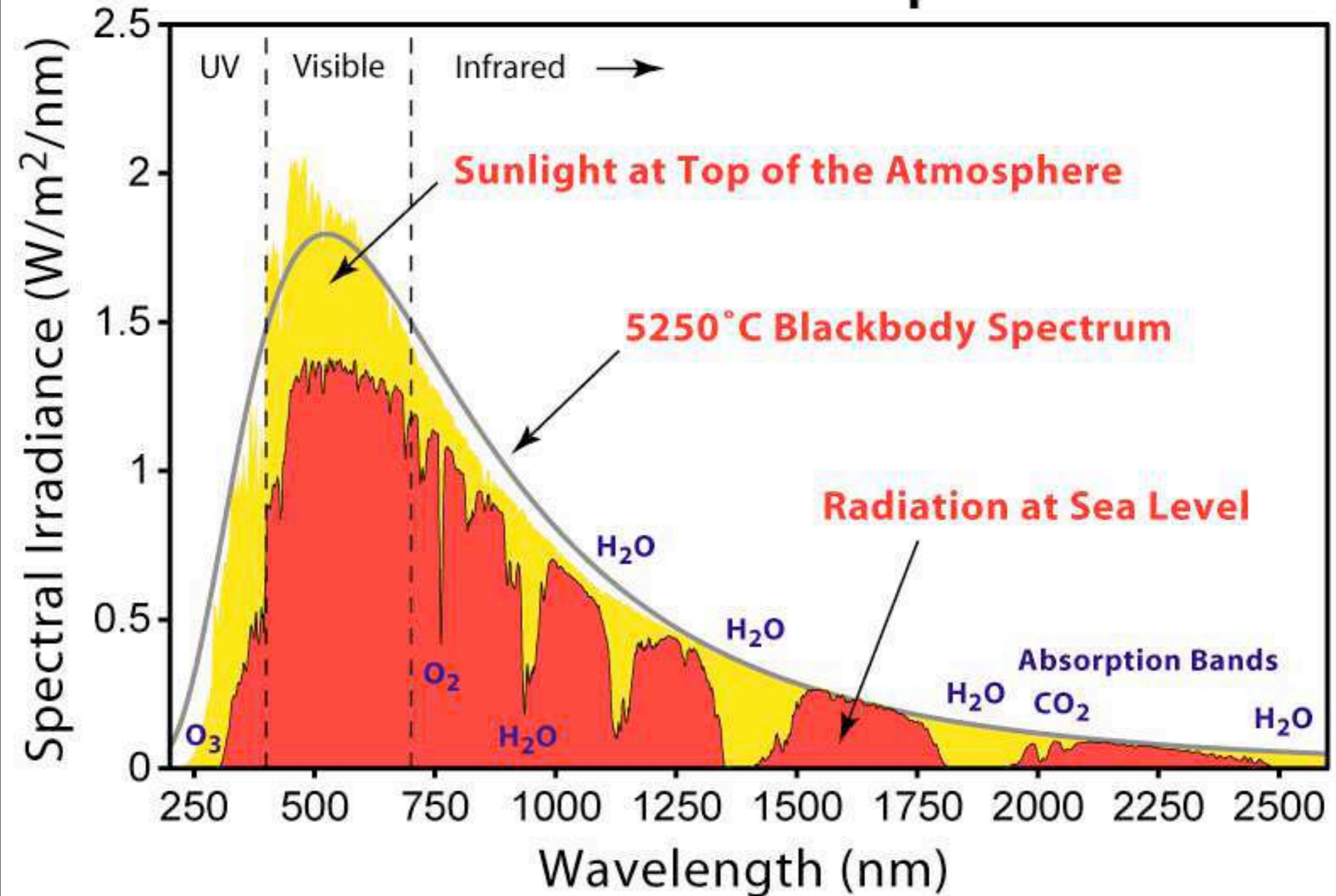


No Star is a Perfect Blackbody

Example: Fraunhofer Lines in the Solar Spectrum



Solar Radiation Spectrum



Multiple Choice Question

Imagine you are sitting in a giant cloud of tenuous gas in outer space. The gas is so tenuous, you can see stars through it, but you can tell that they are a little bit dimmer than they would be if you weren't in the cloud. Suppose the stars are perfect blackbody emitters. If you took a spectrum of any of these stars from inside the cloud, what would you see?

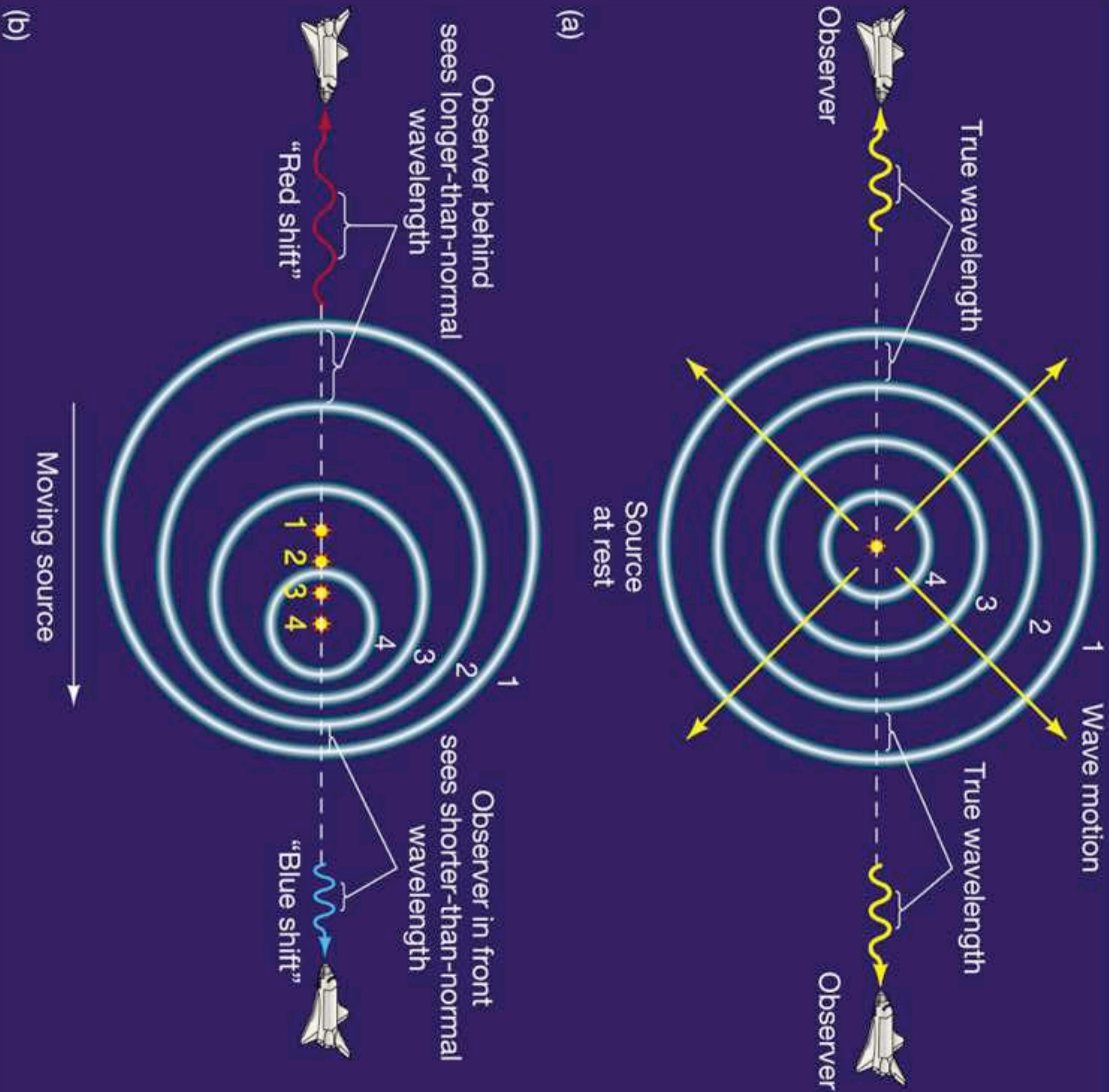
- a) a blackbody or Planck spectrum
- b) emission lines only
- c) absorption lines only
- d) a blackbody spectrum with absorption lines superimposed**
- e) a blackbody spectrum with emission lines superimposed

Doppler Shift

Example: The Pitch of a Train's Horn



Doppler

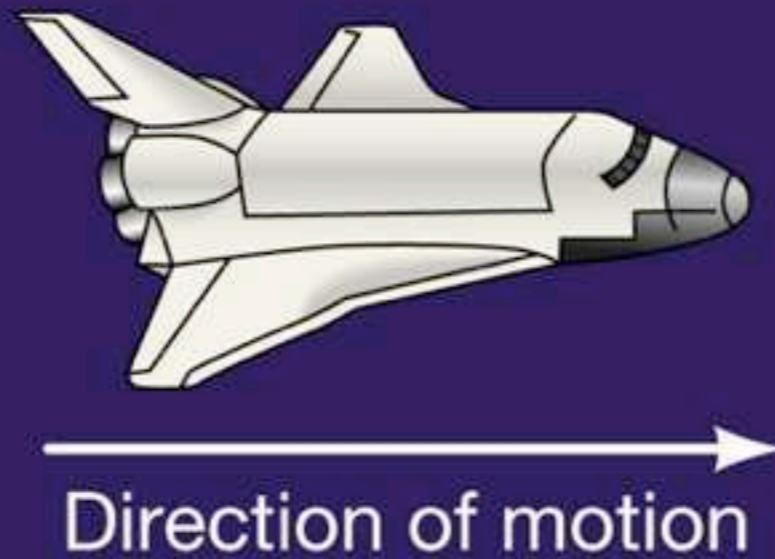


Shift

Doppler Shift (relative motion)

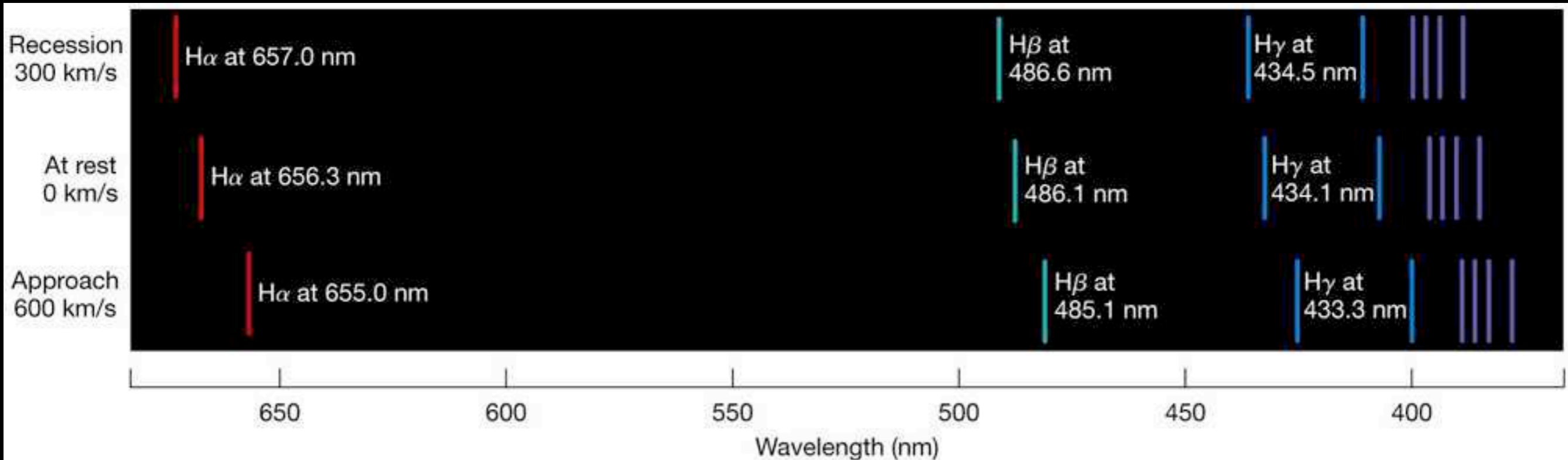
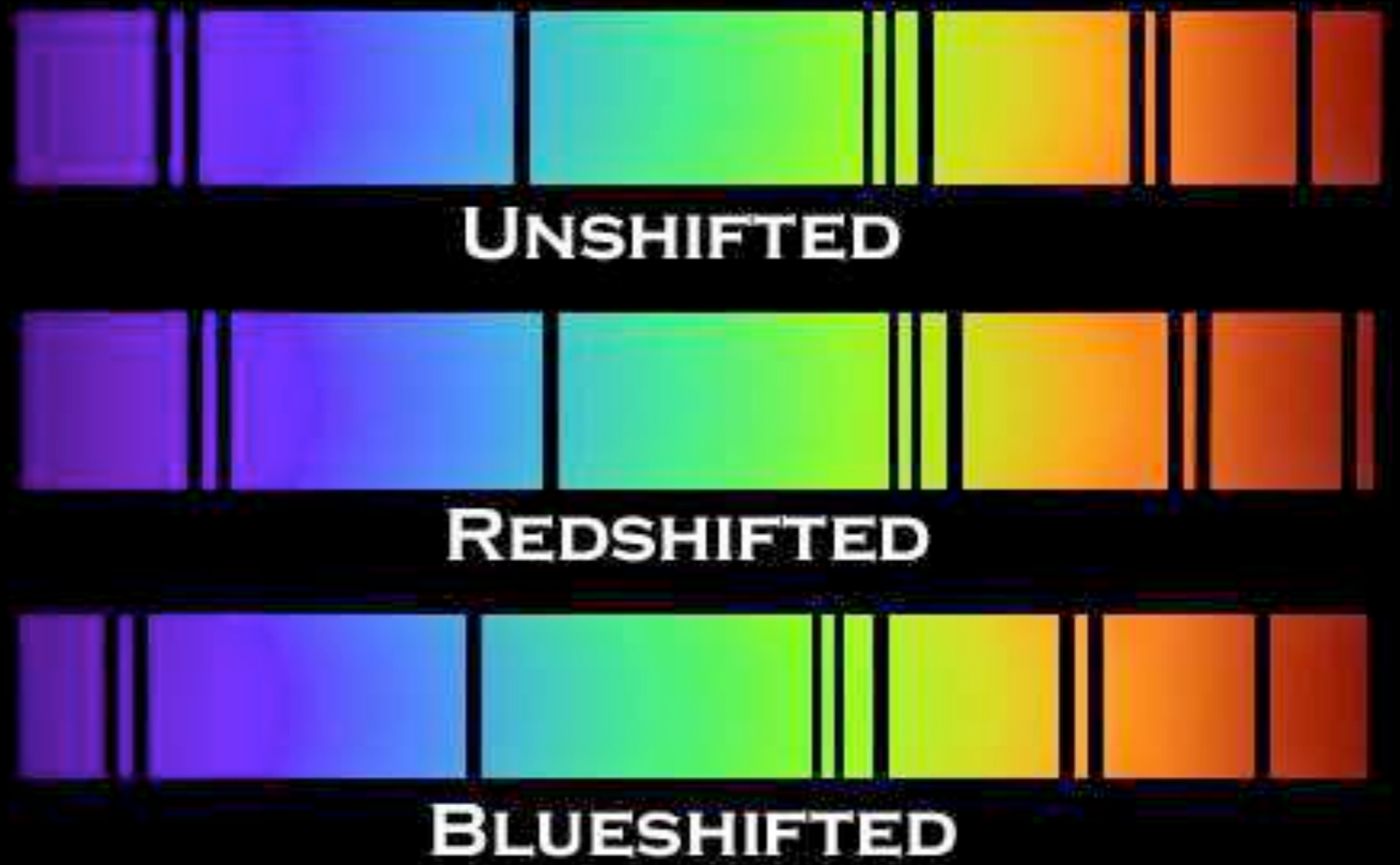
Stars to either
side appear
normal

Stars behind
appear
redshifted



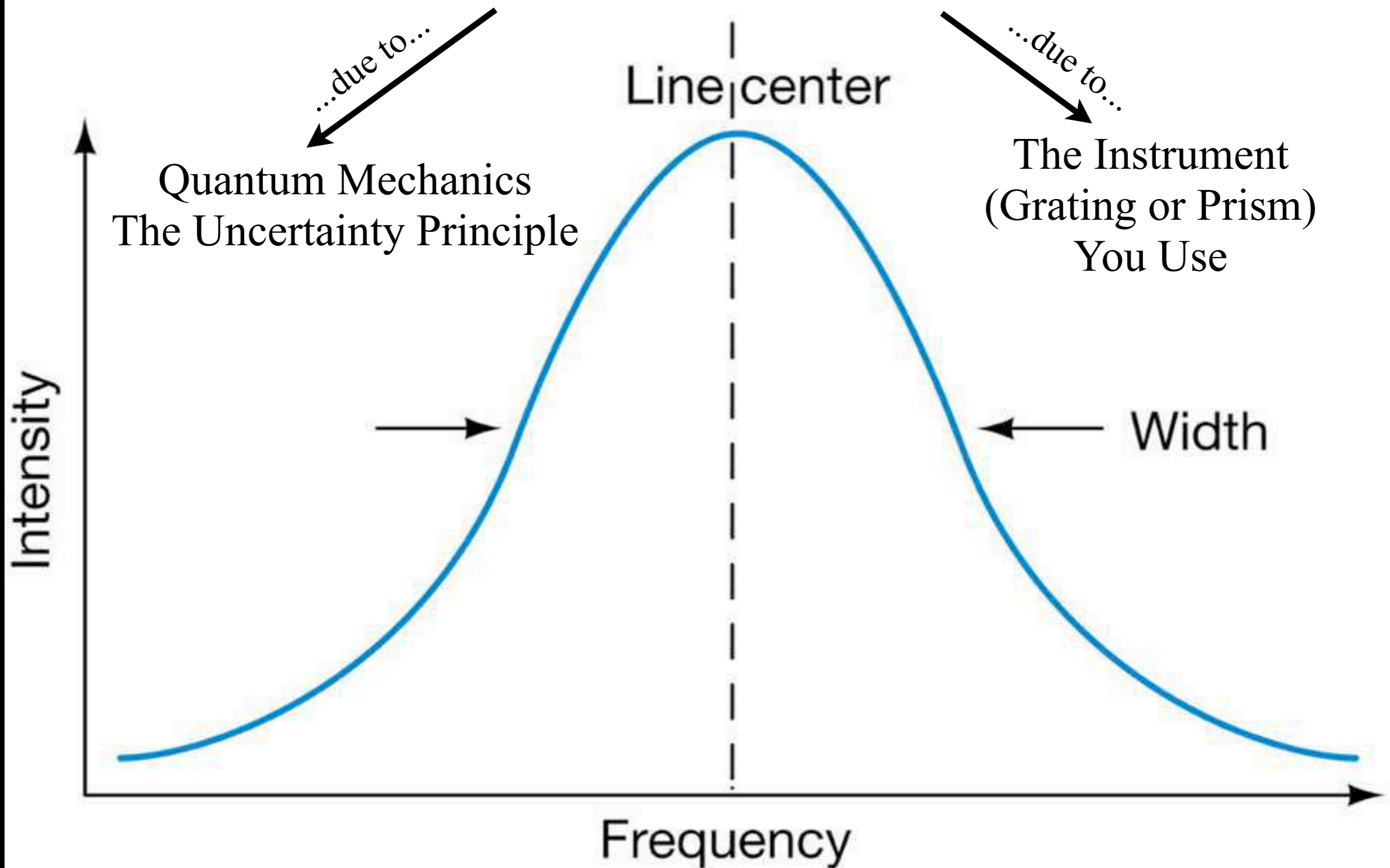
Stars in front
appear
blueshifted

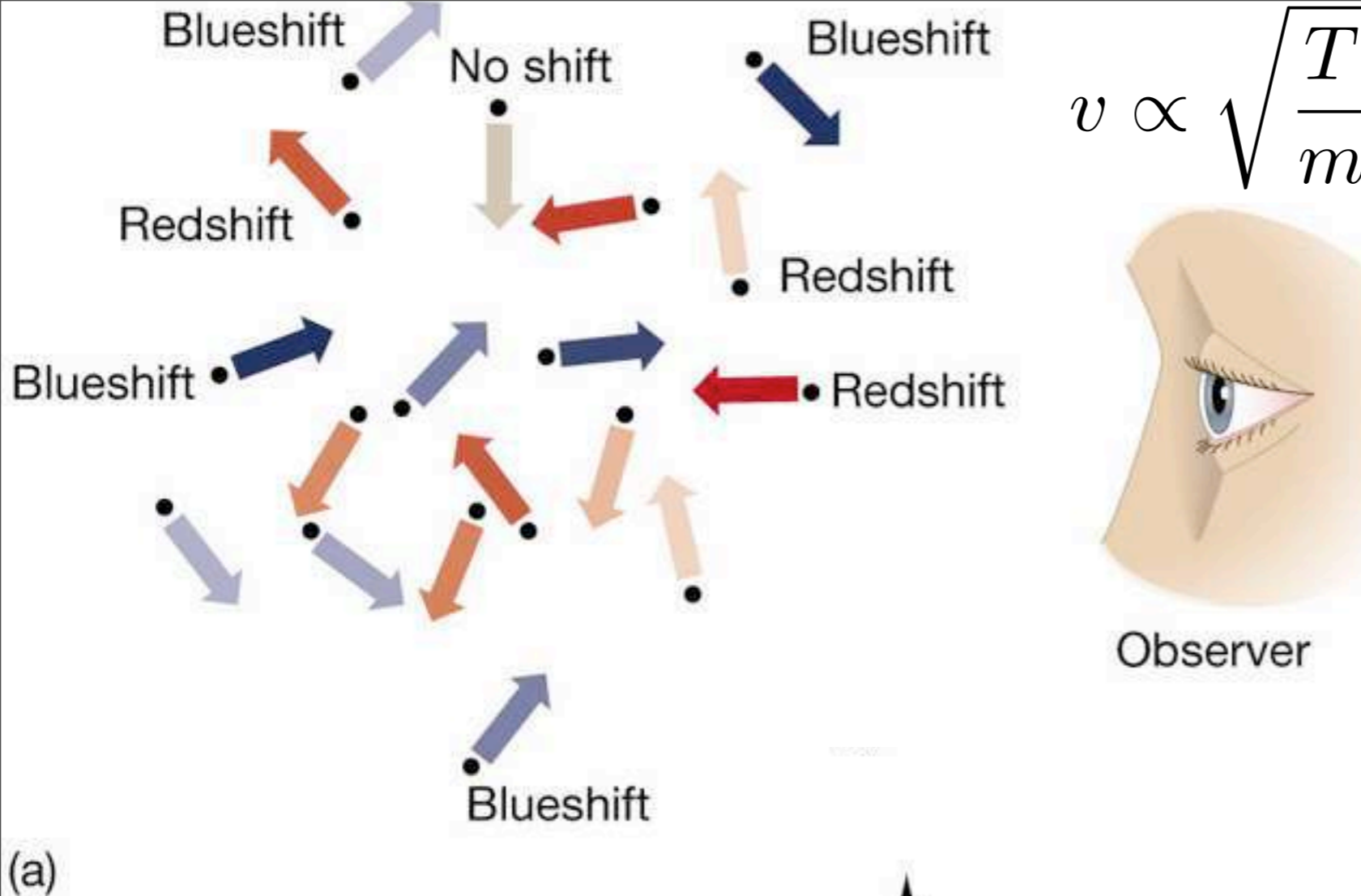
Doppler Shift of Spectral Lines



(a)

Lines Have a Natural Width

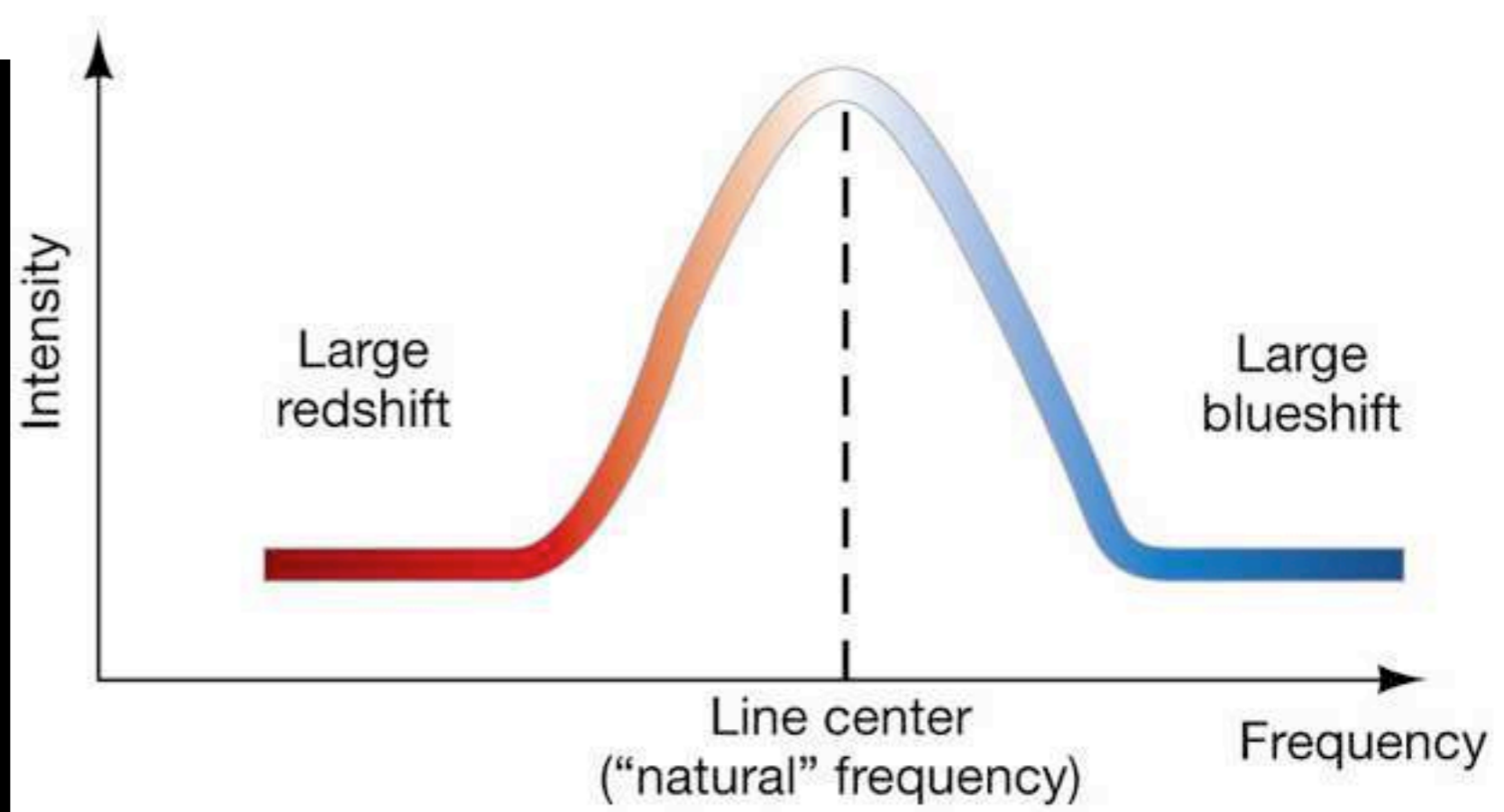




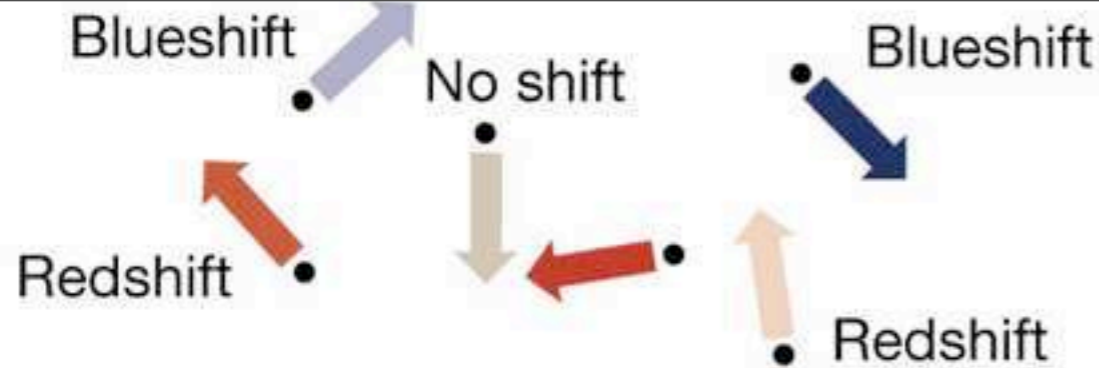
Doppler Broadening

This is why there is essentially no free hydrogen and helium in our atmosphere

(a)



(b)



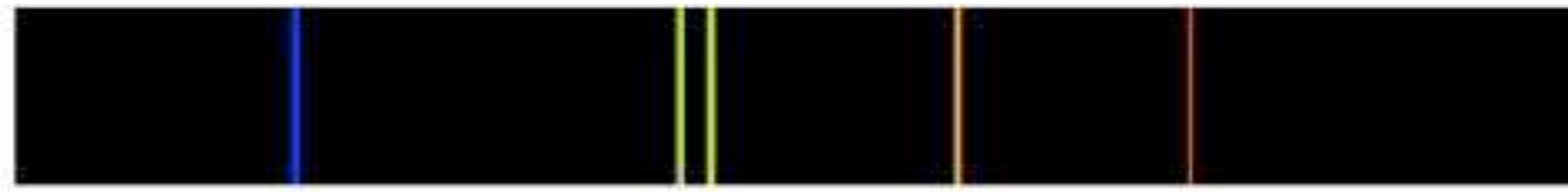
$$v \propto \sqrt{\frac{T}{m}}$$

Doppler Broadening

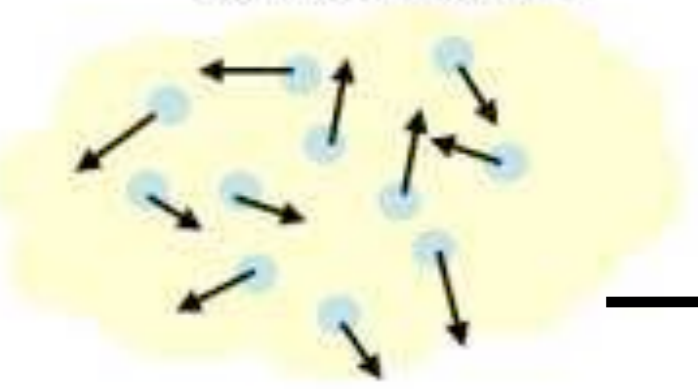
Gas particles at rest



Emission line spectrum with narrow lines



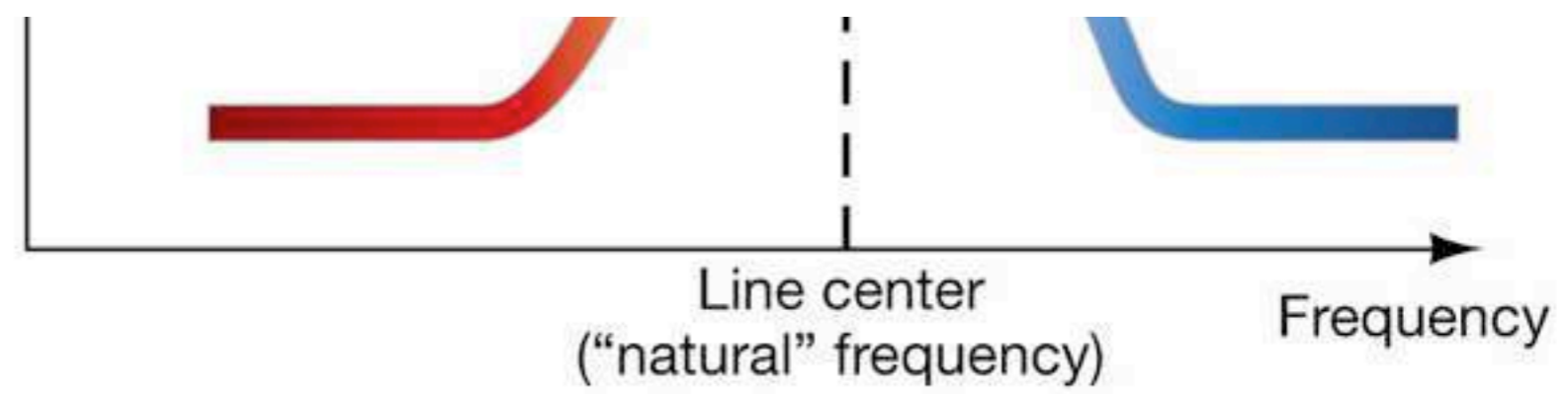
Gas particles with random motions



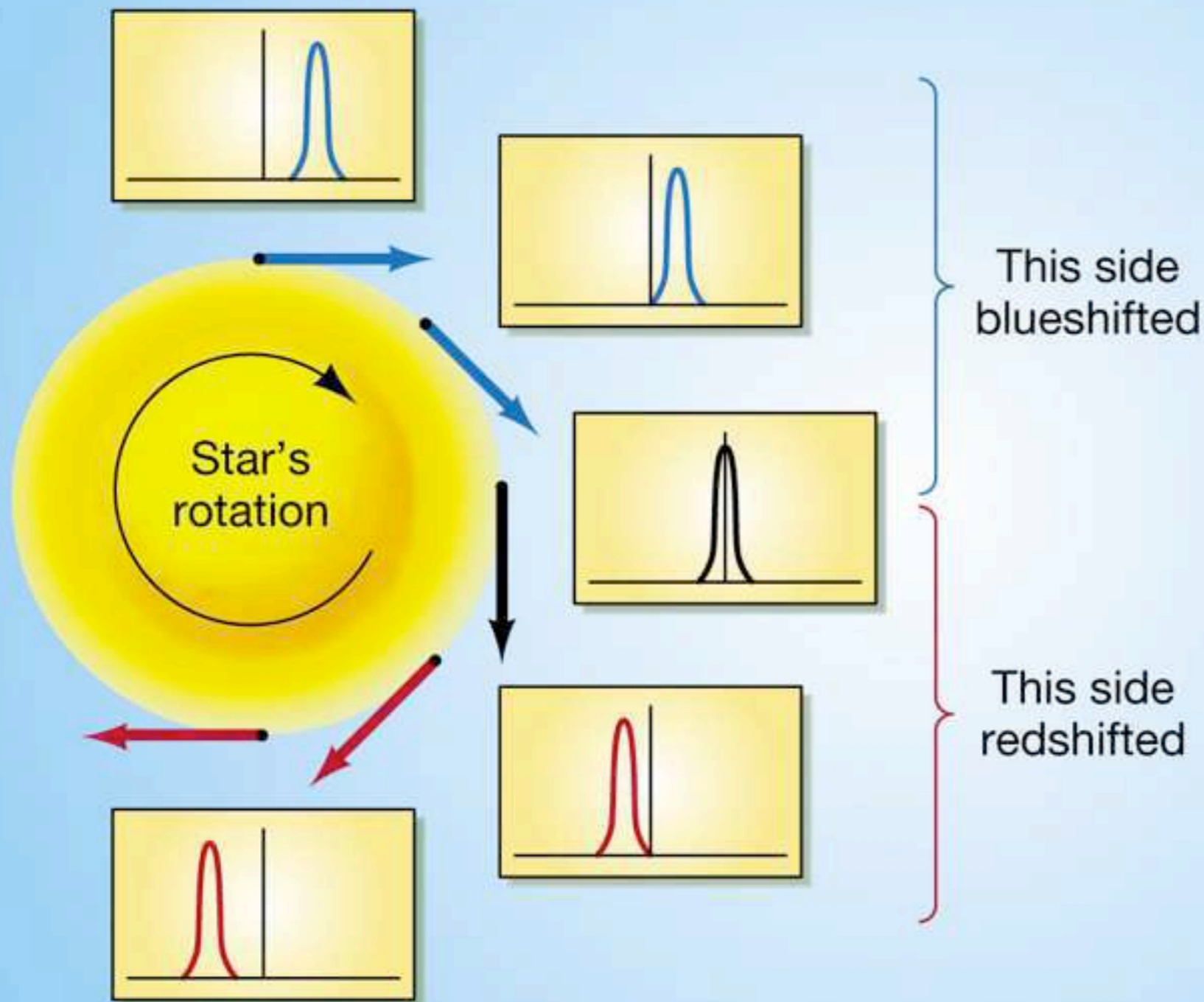
Emission line spectrum with thermal line broadening



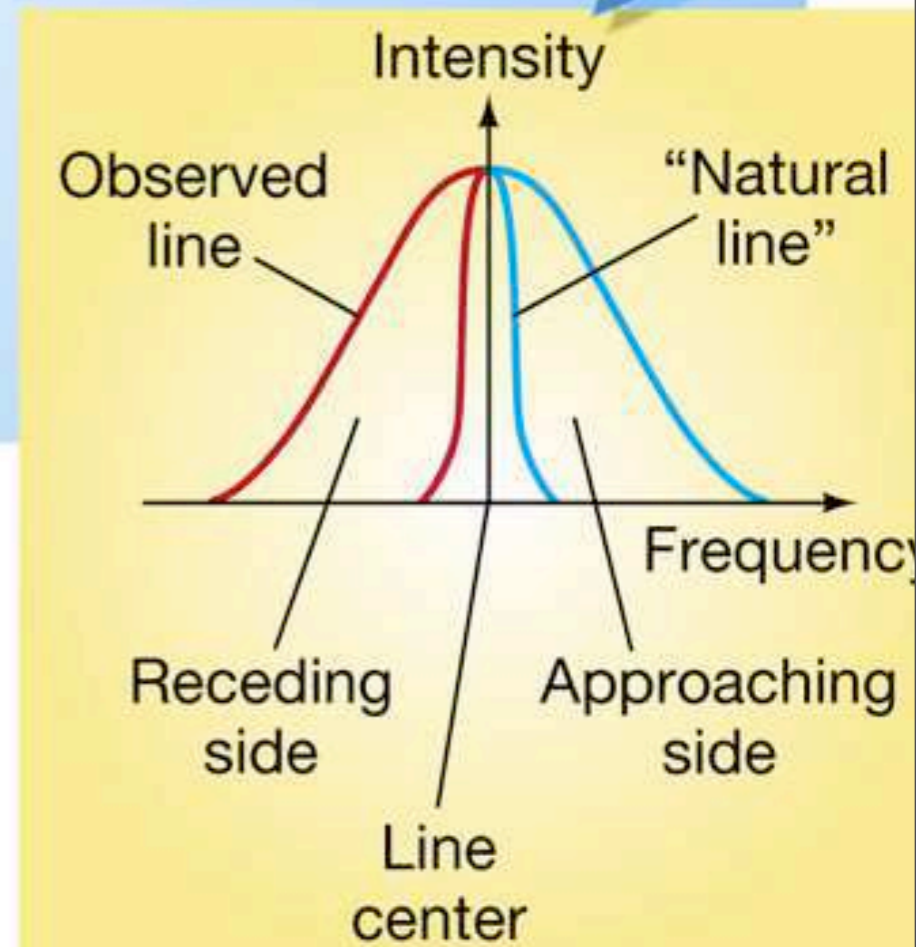
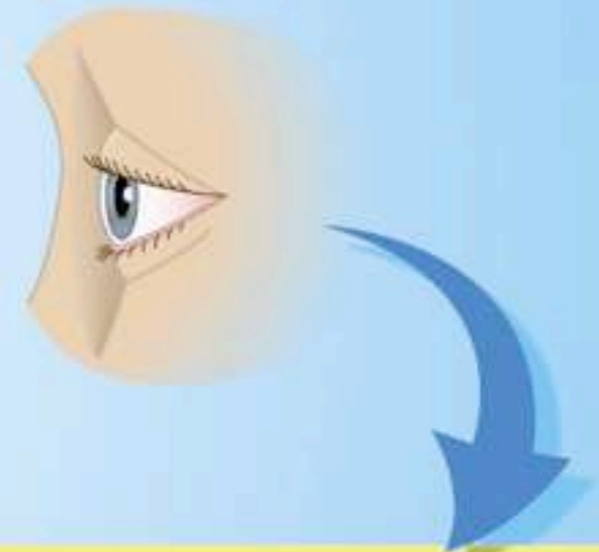
Doppler Broadening + Doppler Shifting is Allowed



(b)



Observer sees:

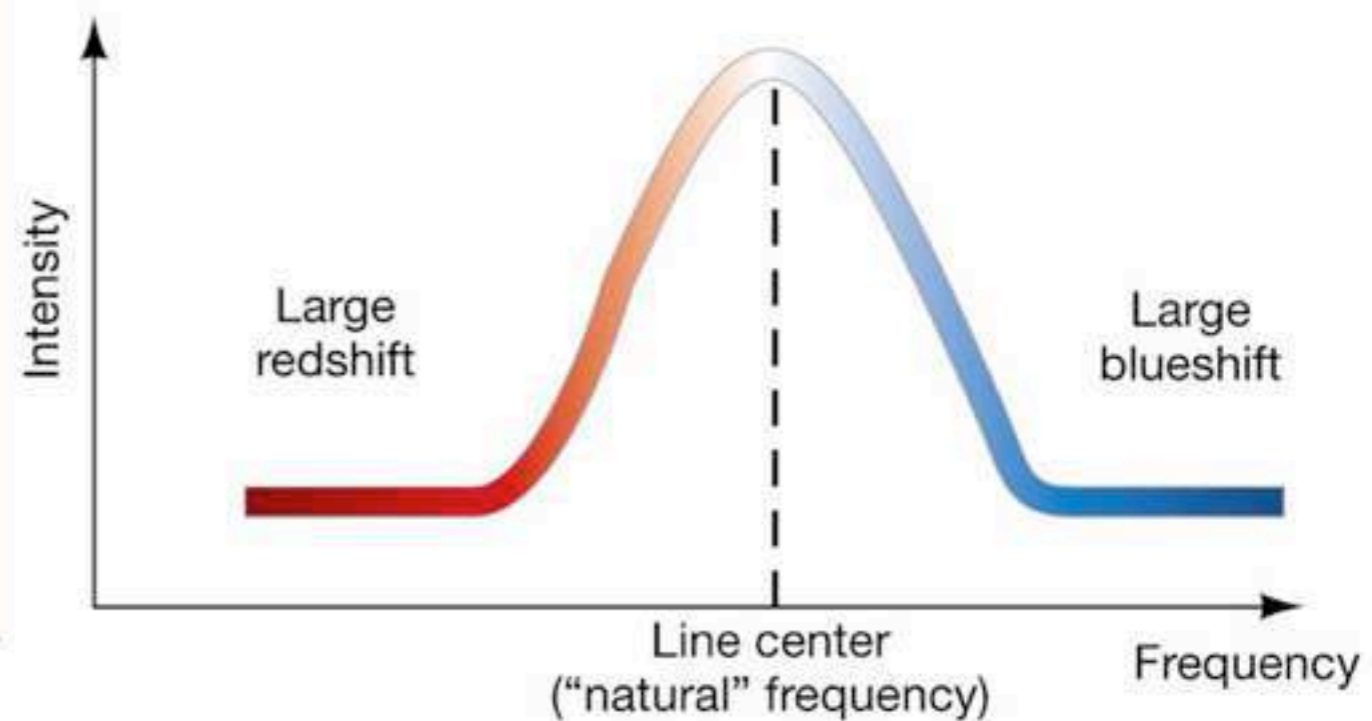
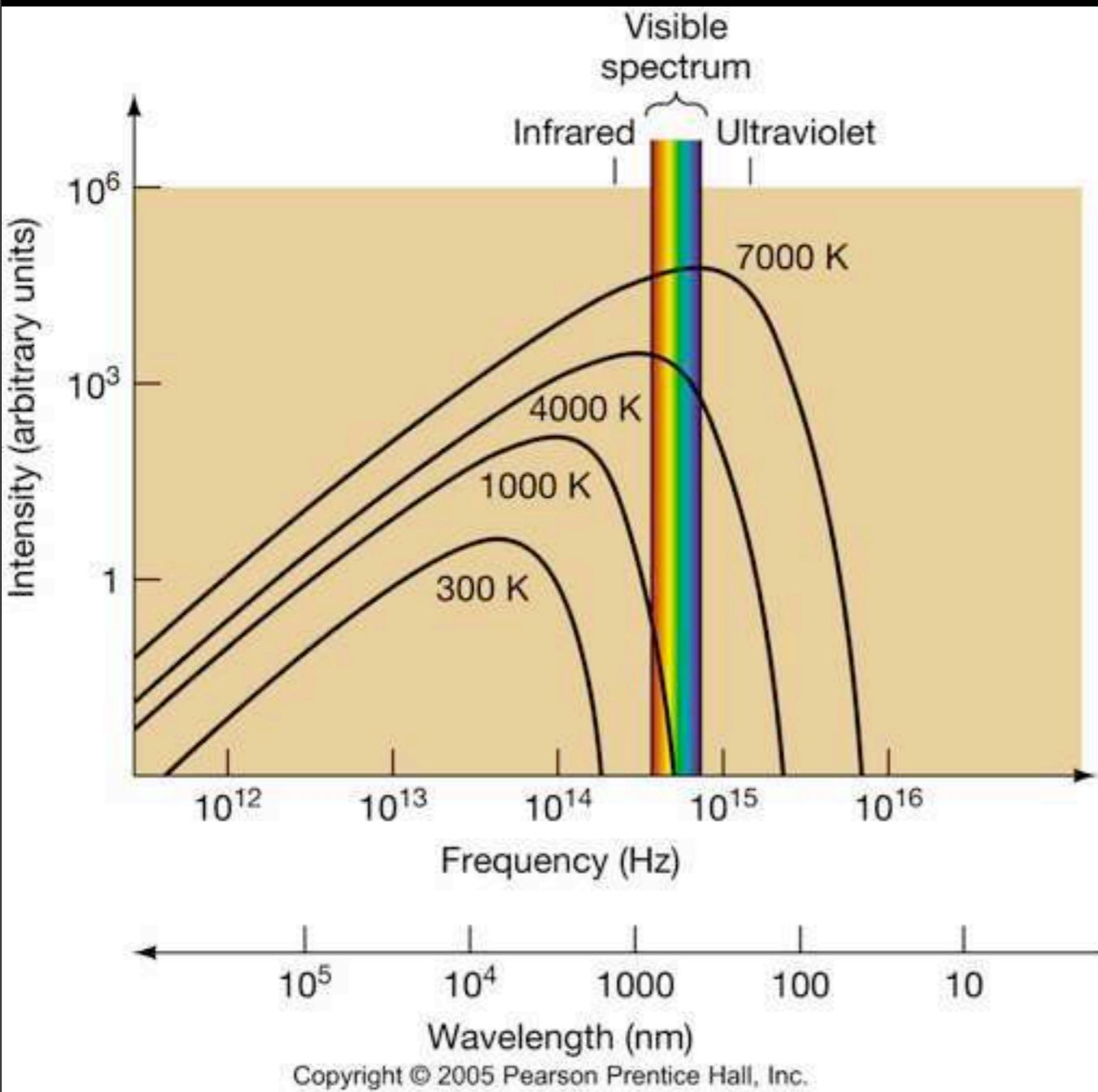


Doppler Broadening

(as seen with stellar rotation)

Why Are Spectra Important?

Temperature



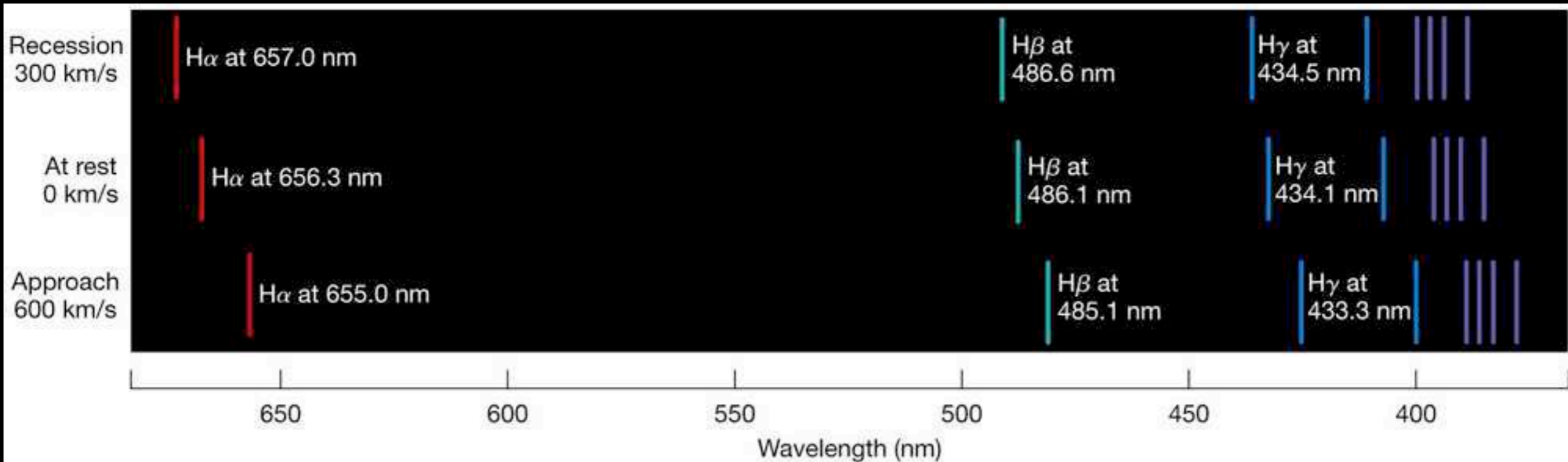
(b)

Discrete Line: Width
(Doppler Broadening)

Continuous Spectrum:
Peak and Flux of Blackbody

Why Are Spectra Important?

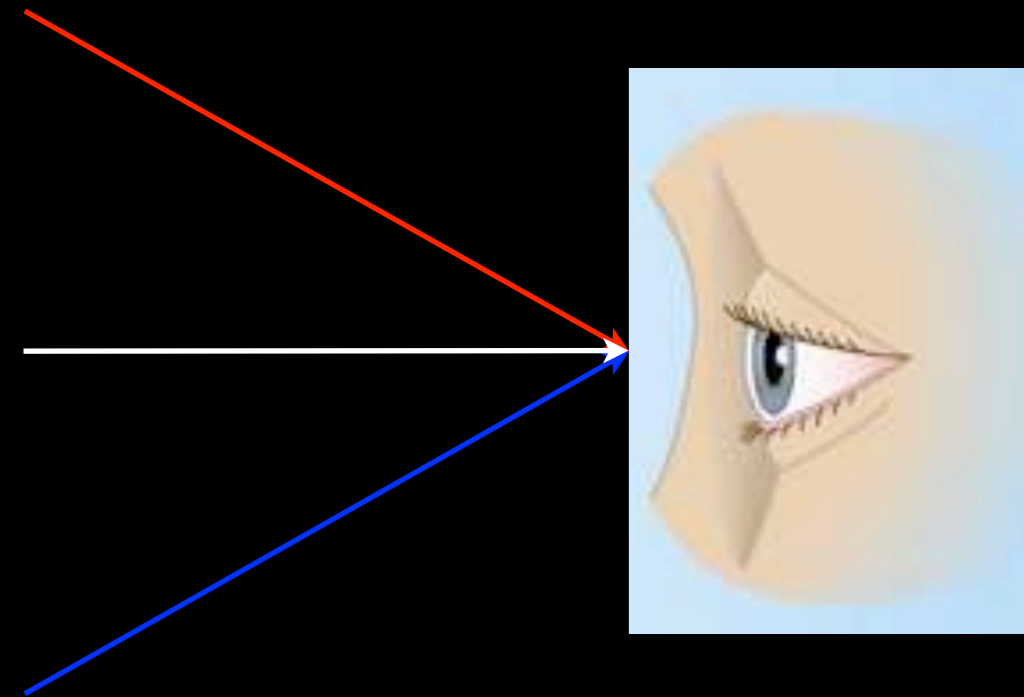
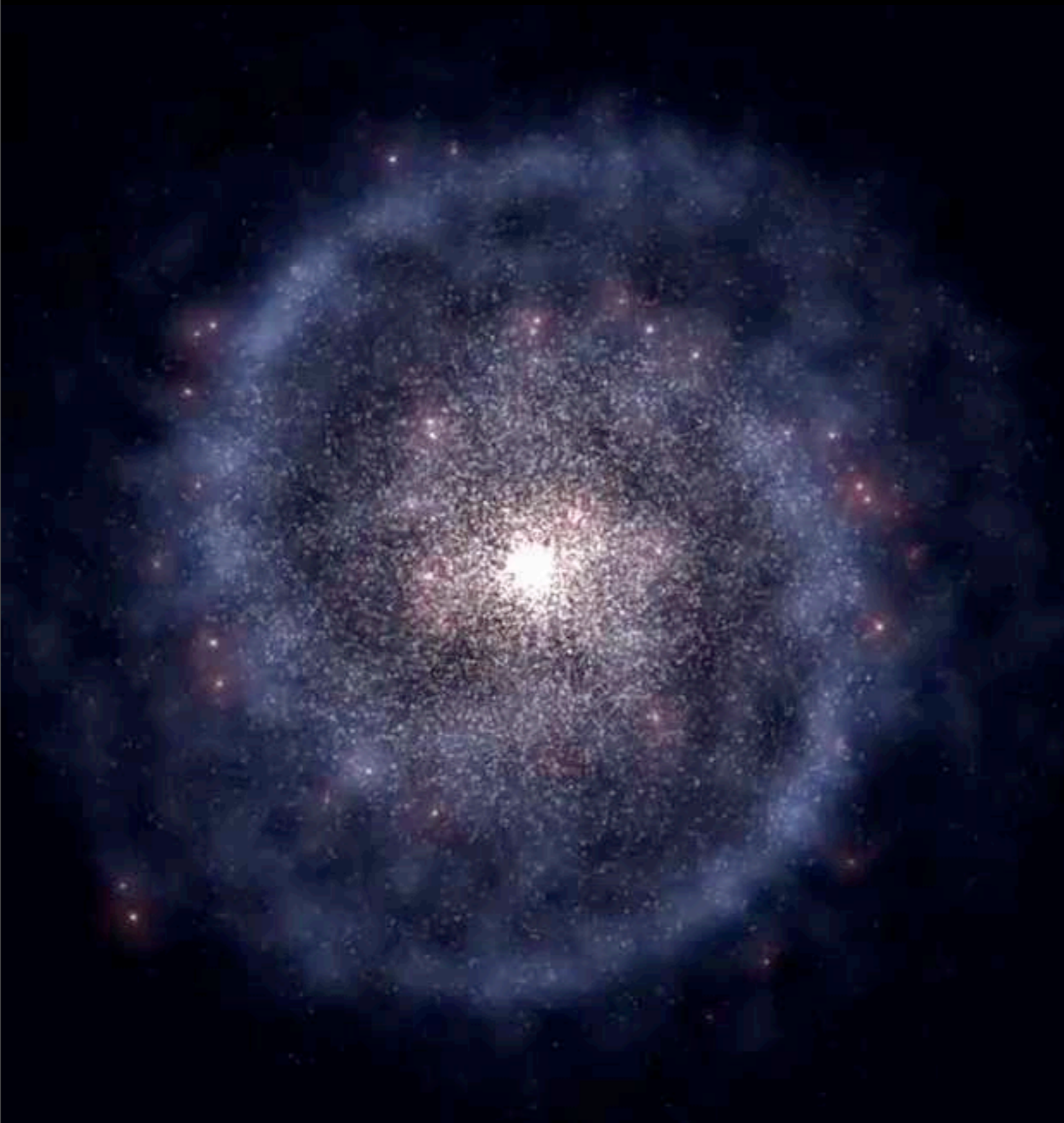
Velocity/Geometry



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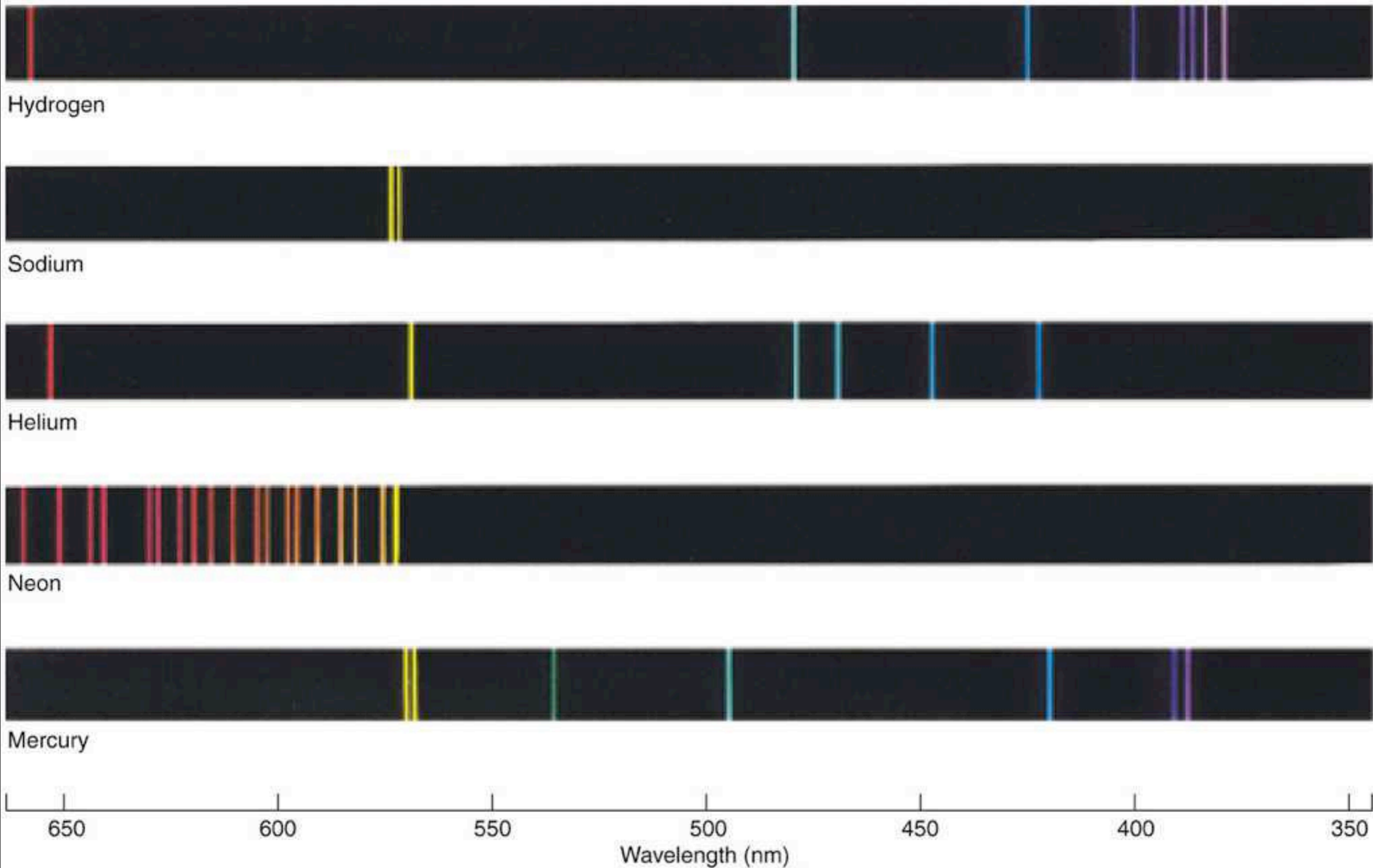
Why Are Spectra Important?

Velocity/Geometry



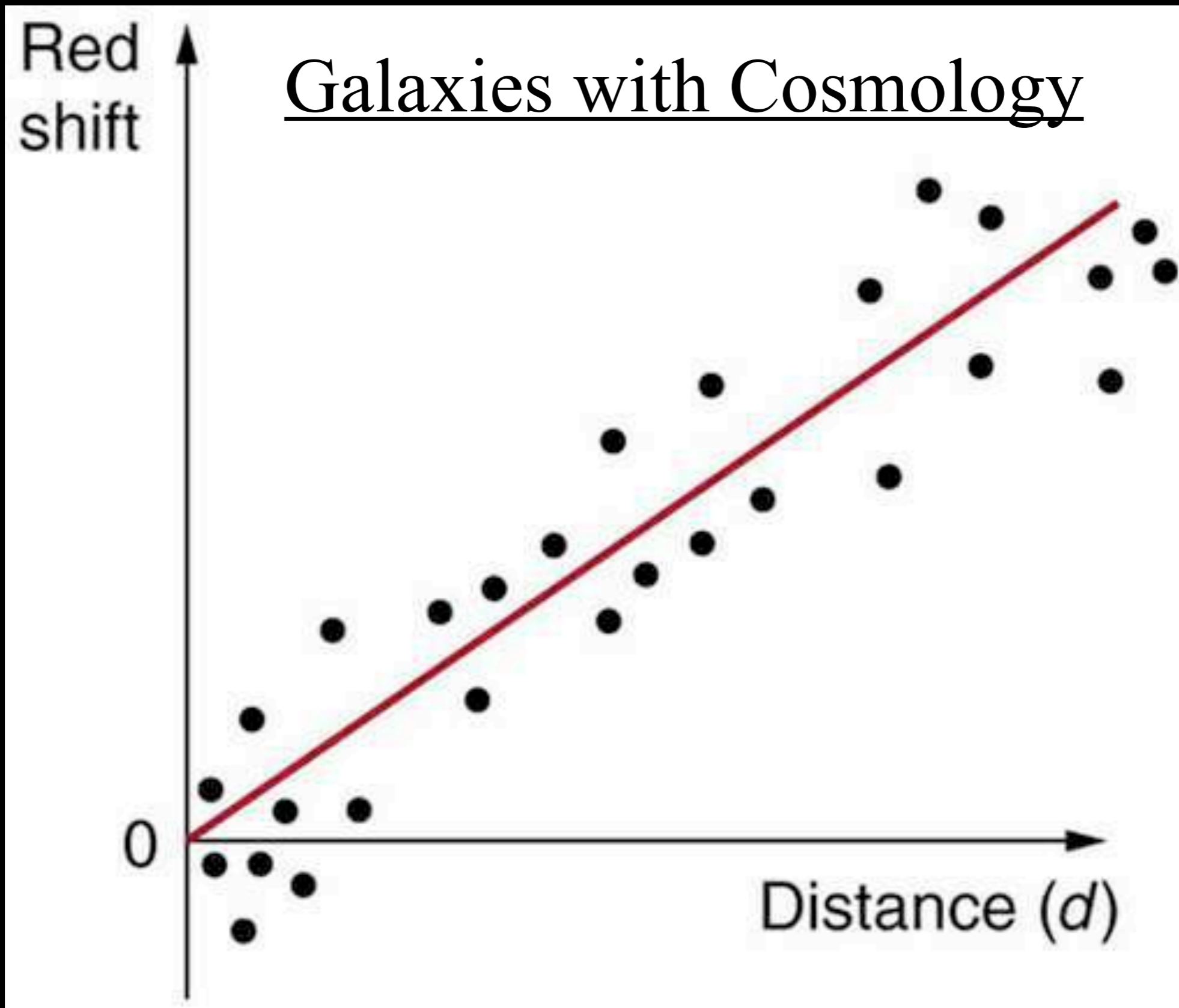
Discrete Line:
Doppler Shift

Why Are Spectra Important? Composition (the element's fingerprint)



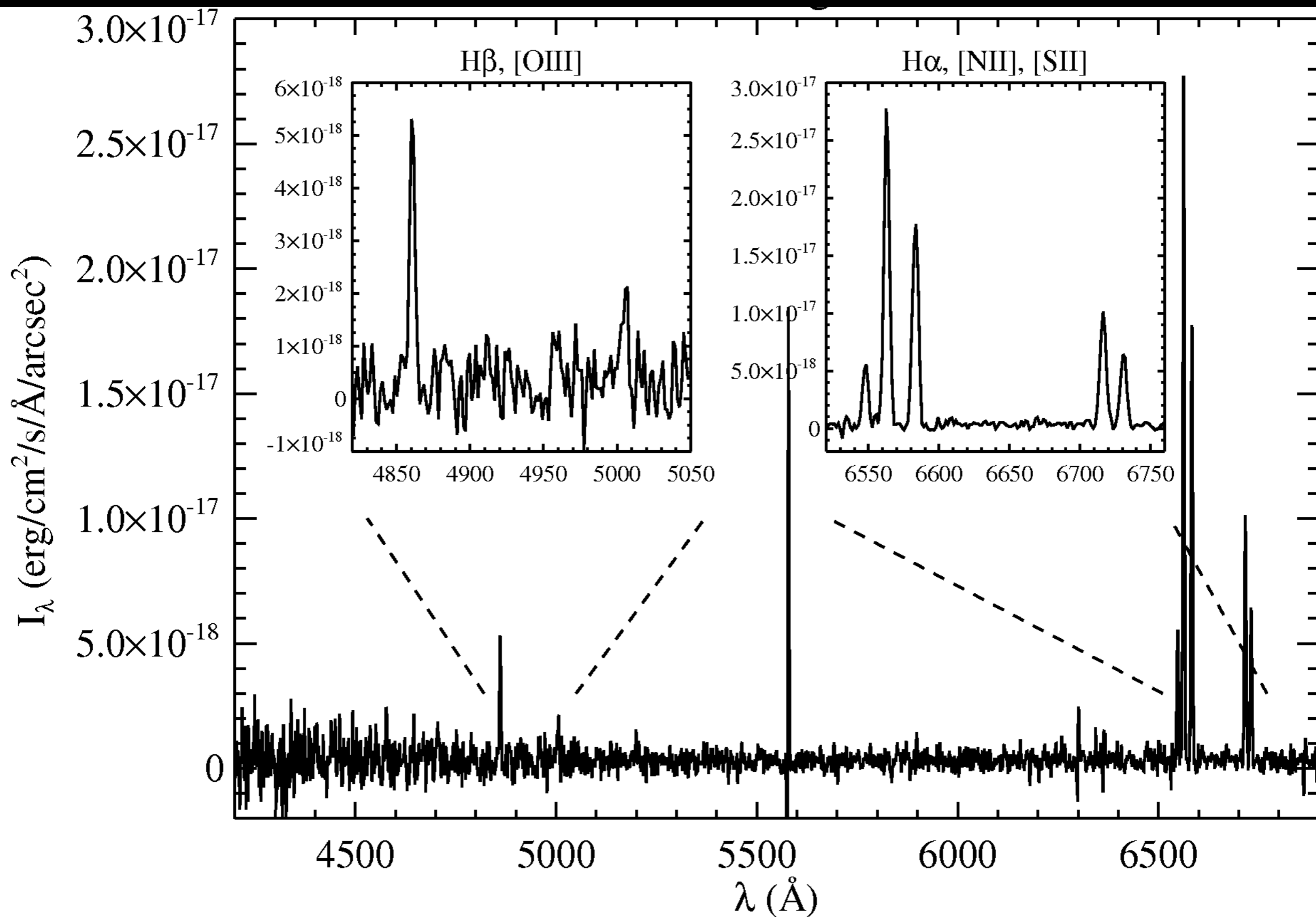
Why Are Spectra Important?

Distance



Why Are Spectra Important?

Temperature and Density



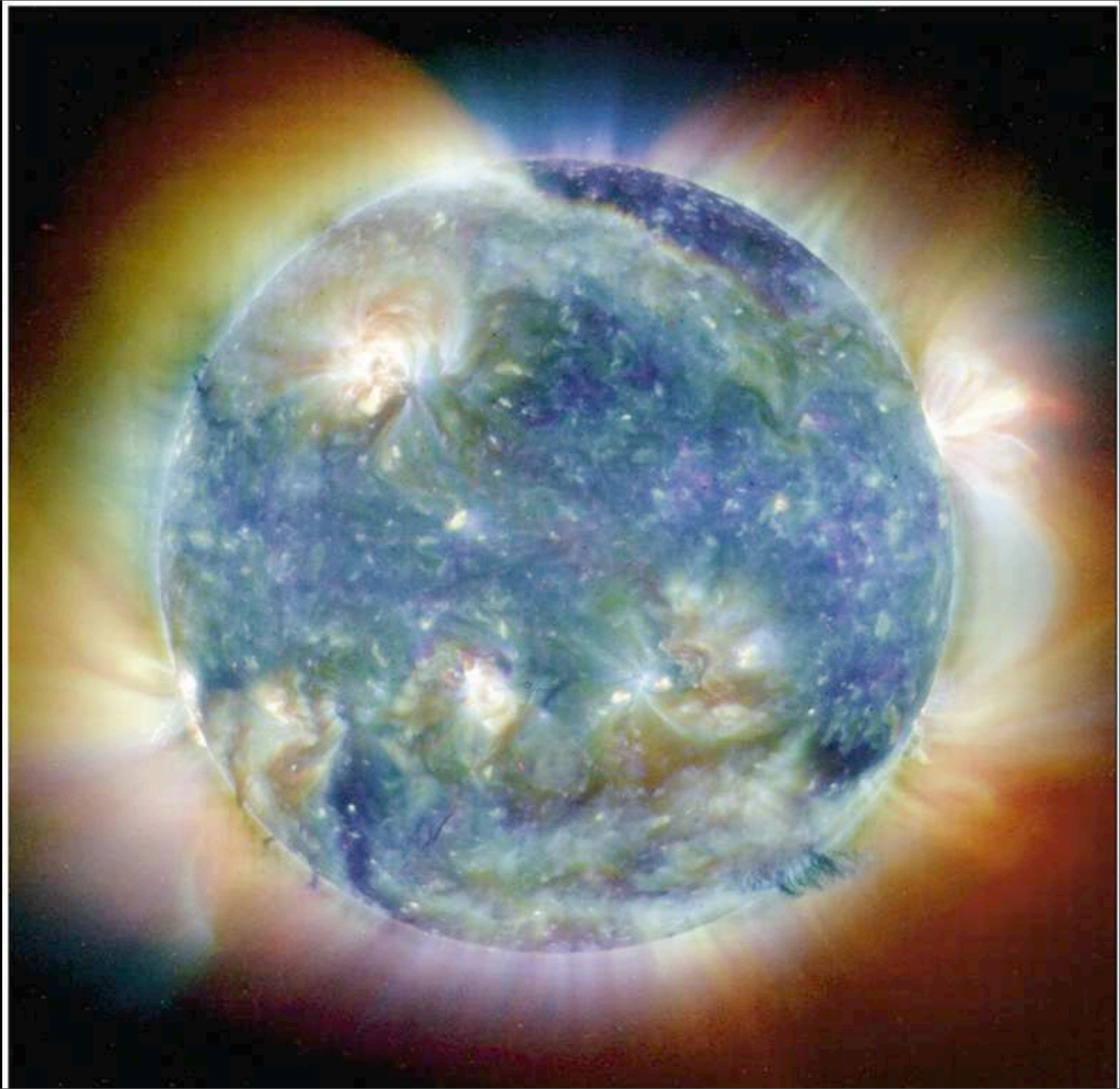
Why Are
Observing at
Different
Wavelengths
Important?

The Sun
(optical)



Why Are
Observing at
Different
Wavelengths
Important?

The Sun
(UV/X-ray)



Multiple Choice Question

You go through a red light and are stopped by a police officer. To get out of being given a ticket you argue: "As I approached the light, officer, it appeared to be green!" It works. Whom of the following should you thank:

Doppler Formula

- a) Kirchhoff (types of spectra)
- b) Wien (blackbody tail and peak shift)
- c) Stefan (flux)
- d) Doppler (shift and broadening)**
- e) Planck (blackbody, quantum theory)

$$\frac{\Delta\lambda}{\lambda} = \frac{v}{c}$$

Relativistic Doppler Formula

$$\frac{\lambda}{\lambda_0} = \sqrt{\frac{1 + v/c}{1 - v/c}}$$

NB: Don't try this! It implies you were traveling at $0.3c$ -- 30% of the speed of light. You will get a very large speeding ticket. Get new brakes.